



ELSEVIER

Contents lists available at ScienceDirect

Data in Brief

journal homepage: www.elsevier.com/locate/dib

Data Article

Sample preparation and electrochemical data of Co_3O_4 working electrode for seawater splittingMalkeshkumar Patel ^a, Wang-Hee Park ^a, Abhijit Ray ^b,
Joondong Kim ^{a,*}, Jung-Ho Lee ^{c,*}^a Photoelectric and Energy Device Application Lab (PEDAL), Department of Electrical Engineering, Incheon National University, 119 Academy Rd. Yeonsu, Incheon 406772, Republic of Korea^b Department of Solar Energy, Pandit Deendayal Petroleum University, Raisan, Gandhinagar 382007, Gujarat, India^c Department of Materials and Chemical Engineering, Hanyang University, Ansan, Kyunggido 426-791, Republic of Korea

ARTICLE INFO

Article history:

Received 18 June 2017

Received in revised form

2 July 2017

Accepted 12 July 2017

Available online 14 July 2017

Keywords:

 Co_3O_4 semitransparent filmPorous Co_3O_4 working electrode

Kirkendall-diffusion

Sea water splitting

ABSTRACT

In this data article, we presented the electrochemical data of the working electrode made of Co_3O_4 semi-transparent film. Electrochemically stable, porous nature of Kirkendall-diffusion grown Co_3O_4 films were applied to generate hydrogen from the seawater splitting (Patel et al., 2017) [1]. The data presented in this article includes the photograph of prepared samples, polarization curves for water oxidation and Tafel plot, linear sweep voltammetry measurements under the pulsed light condition in 0.1 M $\text{Na}_2\text{S}_2\text{O}_3$ electrolyte, and transient photoresponses with natural sea water. Moreover, seawater splitting using the Co_3O_4 working electrode is demonstrated.

© 2017 The Authors. Published by Elsevier Inc. This is an open access article under the CC BY license (<http://creativecommons.org/licenses/by/4.0/>).

* Corresponding authors.

E-mail addresses: joonkim@incheon.ac.kr (J. Kim), jungho@hanyang.ac.kr (J.-H. Lee).

Specifications Table

Subject area	Physics, Chemistry, Electrical Engineering
More specific subject area	Solar Materials, Water Splitting, Hydrogen
Type of data	Photograph, Figures, Video
How data was acquired	Digital camera Potentiostat/Galvanostat (ZIVE, SP1, WonA Tech, Korea)
Data format	Video Clip (.avi) and Analyzed
Experimental factors	Photograph of Sample: Day light Linear sweep voltammetry: <ol style="list-style-type: none"> 1. Polarization curves, scan direction 1.2–2.5 V vs RHE, iR corrected, scan step 5 mV, scan speed 5 mV/s, natural sea water 2. Photocathode, scan direction 0.4 V to –0.7 V vs RHE, scan step 5 mV, scan speed 25 mV/s, electrolyte 0.1 M Na₂S₂O₃ Spectral response: Chronoamperometry technique, Applied potential 0.33 V vs RHE Light source: 365 nm, 2 mW/cm ² 460 nm, 3 mW/cm ² 520 nm, 6 mW/cm ² 620 nm, 15 mW/cm ² Demonstration video: Natural sea water (Yellow Sea near the Incheon National University at 37.3751° N, 126.6328° E coordinate (7 Jun 2016, pH 7.69 (Martini Instruments, pH 56), White light source 100 mW/cm ² , current density 20 mA/cm ² @ –0.8 V vs RHE
Experimental features	Kirkendall diffusion grown porous Co ₃ O ₄ electrode and sea water splitting
Data source location	Incheon National University, Incheon 22012, Korea
Data accessibility	The data are with this article

Value of the data

- Photographs of the Co₃O₄ samples for the large working area of 60 mm².
- Polarization curves for water oxidation can be useful to design the water splitting research.
- Linear sweep voltammetry of Co₃O₄ working electrode in 0.1 M Na₂S₂O₃ electrolyte examined for photocathode properties
- Spectral responses of Co₃O₄/FTO photocathode with natural seawater at over potential of 0.33 V vs RHE are promising for photo-catalyst water splitting application.

1. Data

Fig. 1 shows the photographs of the developed porous Co₃O₄ films on the FTO/glass substrate. Co₃O₄ films were masked using Kapton tape to define the working area (60 mm²) for electro-chemical experimentation are shown in Fig. 1b. Polarization curve for water oxidation cycle (as shown in Fig. 2) are presented for the natural sea water. Tafel analysis is presented in Fig. 2b. Linear sweep voltammetry (LSV) measurement of the porous Co₃O₄ electrode was measured

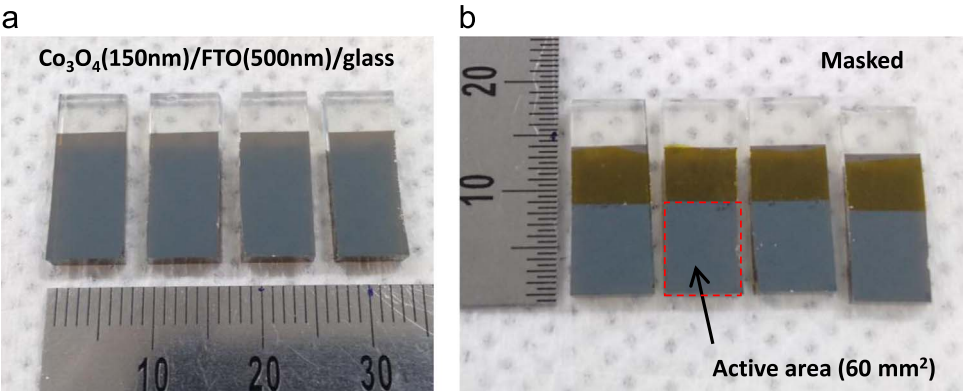


Fig. 1. Photographs of the developed porous Co₃O₄ films on the FTO/glass substrate. (a) as prepared 150 nm thick Co₃O₄ layer on the 500 nm thick FTO-coated glass. (b) Co₃O₄ films were masked using the Kapton tape so that 60 ± 0.5 mm² of the planar area remained for contacting the electrolyte (natural seawater).

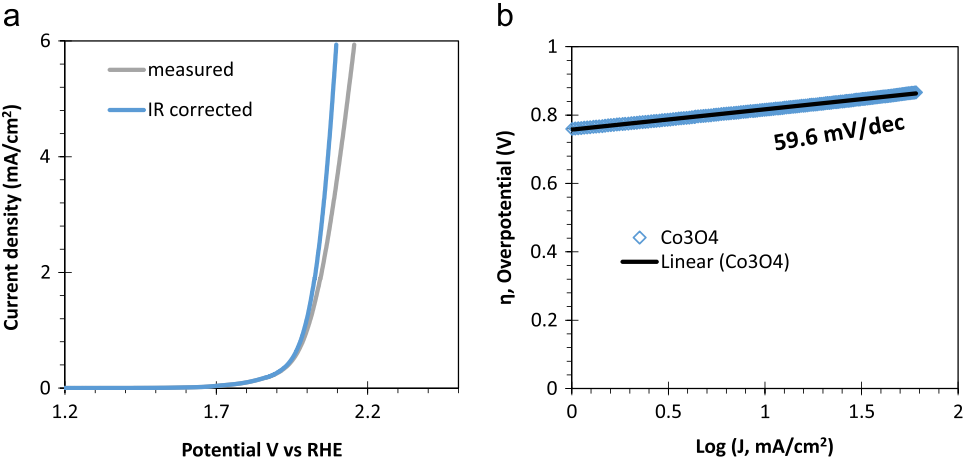


Fig. 2. Electrochemical characterization of the Co₃O₄ electrode for the anodic potential. (a) Current-voltage characteristics and (b) Tafel plot derived from (a).

under a pulsed light (100 mW/cm²) condition with 0.1 M Na₂S₂O₃ electrolyte in cathodic direction, as shown in Fig. 3. Fig. 4 shows the spectral photoresponse of Co₃O₄/FTO photocathode with natural sea water at over potential of 0.3 V vs RHE. Photoinduced seawater splitting using the porous Co₃O₄ working electrode is demonstrated in the moving clip. In this video current density is 20 mA/cm² @ −0.8 V vs RHE.

2. Experimental design, materials and methods

Preparing Co₃O₄ electrode: Co₃O₄ working electrodes were prepared using the Kirkendall diffusion method [1]. Initially the sputtered pure Co film was grown for the porous and semitransparent Co₃O₄ film by the heat treatment (550 °C for 10 minutes in air condition). Photographs of Co₃O₄ samples are shown in Fig. 1.

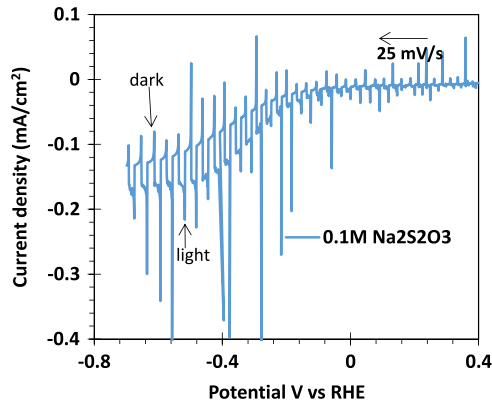


Fig. 3. LSV measurement of porous Co₃O₄ electrodes under a pulsed light condition with 0.1 M Na₂S₂O₃ electrolyte.

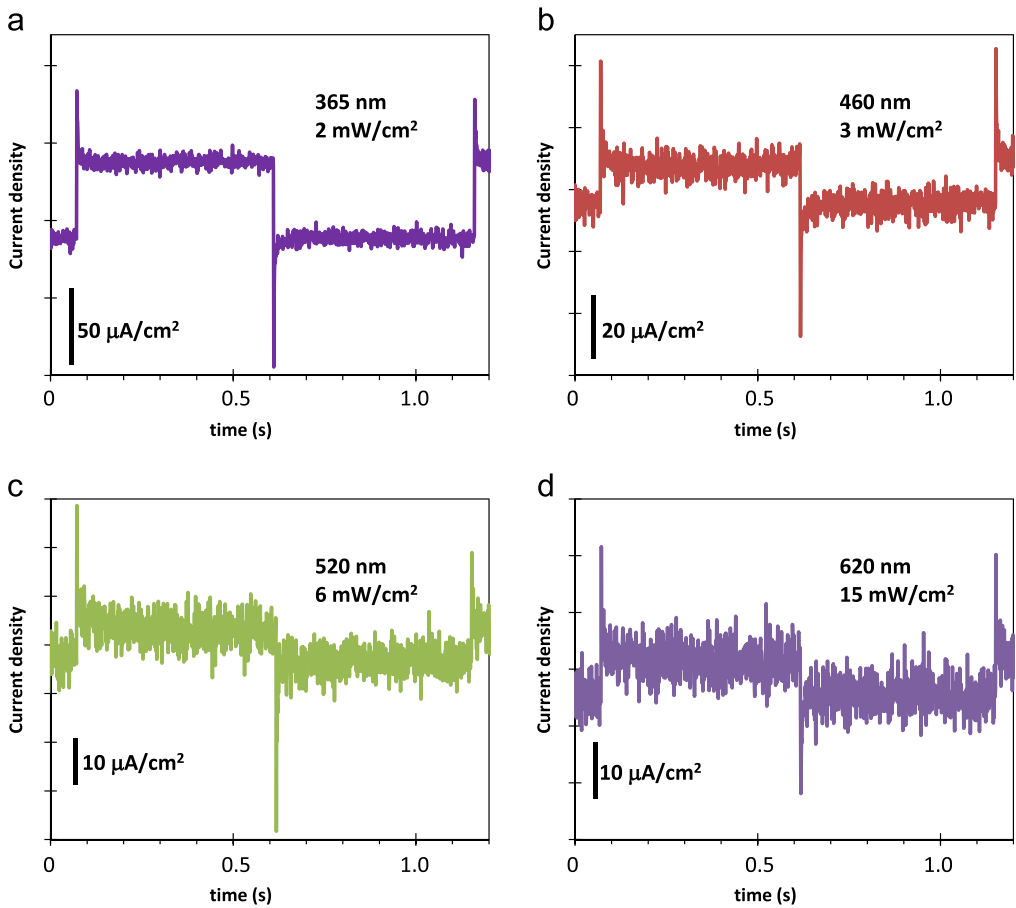


Fig. 4. Spectral response of Co₃O₄/FTO photocathode with natural seawater showed fast transient at applied over potential 0.33 V vs RHE.

Electrolytes:

1. Natural seawater: *Yellow Sea near the Incheon National University at 37.3751° N, 126.6328° E coordinate (7 Jun 2016, pH 7.69 (Martini Instruments, pH 56), White light source 100 mW/cm², current density 20 mA/cm² @ –0.8 V vs RHE*
2. 0.1 M Na₂S₂O₃: Aqueous electrolyte (100 ml) was prepared from the laboratory grade Na₂S₂O₃

Electrochemical Measurements: All measurements were done using the three electrode electrochemical cells (Reference electrode: Ag/AgCl (KCl, 3 M), Counter electrode: platinum gauze, and working electrode: Co₃O₄/FTO/glass) attached to the Potentiostat/Galvanostat (PG-stat) (WonA Tech, ZIVE SP1). Linear sweep voltammetry was applied to measure the anodic polarization and photocathode properties. Chronoamperometry technique was applied to measure the transient photoresponses at an applied potential of 0.33 V vs RHE in the natural seawater. The white light (5800 K, Bridgelux, ES Star Array, BXRA-56C0700-A) was applied for photoinduced electrochemical measurement. This was calibrated by a power meter (KUSAMMECO, KM-SPM-11). The illuminating light source was calibrated for one-sun light intensity (100 mW/cm²) and was applied in the pulse mode or the continuous mode. For transient photoresponses, a monochromatic light source of wavelength 365 nm (2 mW/cm²), 460 nm (3 mW/cm²), 520 nm (6 mW/cm²), and 620 nm (15 mW/cm²) were applied to working electrode from the front direction.

Acknowledgements

This research was supported by Basic Science Research Program through the National Research Foundation (NRF) of Korea by the Ministry of Education (NRF-2015R1D1A1A01059165), Korea Research Fellowship Program through the NRF by the Ministry of Science, ICT and Future Planning (NRF-2015H1D3A1066311) and Korea Institute of Energy Technology Evaluation and Planning by the Ministry of Knowledge Economy (KETEP-20168520011370)

Transparency document. Supplementary material

Transparency document associated with this article can be found in the online version at <http://dx.doi.org/10.1016/j.dib.2017.07.030>.

Appendix A. Supplementary material

Supplementary material associated with this article can be found in the online version at <http://dx.doi.org/10.1016/j.dib.2017.07.030>.

Reference

- [1] M. Patel, W.H. Park, A. Ray, J. Kim, J.H. Lee, Photoelectrocatalytic Sea water splitting using Kirkendall diffusion grown functional Co₃O₄ film, *Sol. Energy Mater. Sol. Cells* (2017) (In process (minor revision)).