





Review

The Environmental Significance of Contaminants of Concern in the Soil–Vegetable Interface: Sources, Accumulation, Health Risks, and Mitigation through Biochar

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Abstract: Vegetable consumption is considered as an important part of the human diet as it serves as an essential source of vitamins, nutrients, and minerals. In this regard, the demand for new technologies and ideas in the agricultural sector has grown steadily to help expand the production of vegetable crops. The uptake and accumulation of trace elements (TEs) and pharmaceuticals and personal care products (PPCPs) as contaminants in vegetables have been accelerated by man-made activities. The dietary intake of these contaminated vegetables often poses significant human health risks. To counteract this, mitigation strategies in the form of environmental amendments have received increasing attention in the last decade. The incorporation of amendments in the form of biochar has been shown to reduce the uptake of contaminants in the soil and their accumulation in vegetables. The present review is organized to offer an overview of the occurrence and sources of important contaminants of concern particularly associated with vegetable plants. The factors influencing their uptake and accumulation in the edible parts of vegetable plants are discussed briefly along with the human health risk imposed via the consumption of contaminated vegetables. Furthermore, this review also explores feasible mitigation strategies through the use of biochar for these contaminants, along with future perspectives for addressing this issue of food contamination.

Keywords: trace elements; pharmaceuticals and personal care products; vegetable; uptake; human health; biochar



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1. Introduction

The contamination of soil and agricultural produce has become a serious problem for food security and the environment due to an increase in anthropogenic activities [1]. Recent studies have reported an exponential rise in vegetable consumption in the urban population [2,3]. These studies also emphasized that vegetables grown in urban areas were exposed to higher contaminants than those grown in rural areas. A large fraction of such contamination can be accounted for by the presence of well-known traditional pollutants, namely trace elements (TEs), (such as Cd, Cu, Pb, As, Si, Co, Hg, and Cr) and new emerging contaminants (ECs), such as synthetic chemicals (e.g., pharmaceuticals and personal care products (PPCPs), cosmetics, industrial solvents, artificial sweeteners, microplastics, polycyclic aromatic hydrocarbons, engineered nanomaterials, perfluorinated chemicals, and pesticides) in the food chain. Among these pollutants, the uptake of TEs in the soil–plant system has been reported in various studies [2,4–6]. Moreover, in recent years, the accumulation of ECs (the most prominent being PPCPs in vegetables)

have also attracted the attention of various researchers [7–9]. The major sources of such rapid contamination include industrial activities, wastewater irrigation, waste disposal, or manure amendments [10]. Reusing reclaimed wastewater for irrigation and agricultural practices through treatment plants has led to the deposition of such contaminants in the soil [11]. These contaminants are transferred from the soil via the plant roots and then later accumulate in the edible tissues of vegetable plants [12]. In this regard, the consumption of TEs and PPCPs through contaminated vegetables may pose potential health risks, such as gastrointestinal disorders, cancer, central nervous system disorders, and kidney or liver disorders in the human body [13]. In order to reduce the risk effect of such contaminants of concern several mitigation strategies are often applied in the soil. Adding amendments is one of the common mitigation methods gaining interest in the last decade. Biochar is one such amendment that is receiving increasing attention more recently. Biochar is known to improve soil fertility and reduce the uptake of contaminants from the soil [14]. It is formed by pyrolysis of any biomass in an anaerobic condition [15]. Biochar is known to improve the soil physical, chemical, and biological characteristics as well as reduce the bioavailability of pollutants in vegetables leading to a decrease in risks for humans [16]. Biochar contains a high cation exchange capacity, a large surface area and pore size, an excellent aromatic structure, carbon content, and functional groups which help in the efficient adsorption of TEs and PPCPs, as found in recent studies [17–20].

The contamination of agricultural products is a concern for our food safety and human health. Although many review articles have already presented the uptake and accumulation of TEs in plants, few reviews on the co-occurrence on the uptake of TEs and PPCPs, particularly in vegetables, exist [3,13]. Therefore, the occurrence of mixed contaminants in food has become a research interest. Moreover, the database for PPCPs accumulation in the soil–vegetable system is not sufficient enough to develop proper protections and conclusions about their effects on human health through the consumption of contaminated food. In the light of this limitation, it is worth discussing and collating the information about the potential uptake and accumulation of various soil contaminants especially in the food contained in the daily food baskets of common people. Therefore, the present review aims to give a brief overview of the contaminants of concern in the soil–vegetable interface.

This review initially focuses on the outline of TEs and PPCPs, mainly determining the uptake in vegetable plants. It then elaborates on the various sources and factors influencing the uptake of contaminants. In addition, the review highlights the concentration levels and accumulation patterns of TEs and PPCPs in the edible parts of vegetables and their associated health risk through consumption. Moreover, the recent mitigation strategies through biochar are described along with future perspectives in this research field.

2. Contaminants of Soil–Vegetable Interface

2.1. TEs

Vegetables contain essential minerals that comprise an important part of the human diet [21]. Minerals contain groups of metals and non-metals that play an important role in the growth and metabolism of plants. They can be categorized into macro/major or micro/trace elements [13]. Macro-elements, such as Ca, Mg, N, and P, are required in plants at high concentrations (>0.1% dry weight). In contrast, TEs are required in low amounts in plants and can be further subdivided into essential and non-essential TEs. Zn, Mo, Fe, Ni, B, Cr, and Cu are examples of essential TEs required for plant metabolism and development [22]. However, as the concentration of essential TEs exceeds 0.1% of dry weight, they can turn into toxic and harmful substances in the environment.

In contrast, non-essential TEs, such as Cd, As, Pb, and F, are not required for plant growth or development and are highly toxic. High concentrations of TEs can affect soil, plants, food crops, human health, and the surrounding environment in a negative manner [23]. TEs, such as Cd, Pb, As, Cu, and Cr, tend to exhibit high accumulation rates in the soil–vegetable interface. Such TEs are translocated in much greater amounts in

plants and vegetables due to similar physio-chemical properties [24,25]. According to the United States Environmental Protection Agency, Pb is an extremely toxic metal present in the environment [26]. Noxious concentrations of Pb can inhibit the enzymatic activities of plants while decreasing the total protein content required for proper functioning in plant tissues [27]. Furthermore, Cd toxicity can decrease the seed germination process and affect the nutrient content in plants [28]. Pb and Cd can show intense mobility and translocation rates from the soil to plant roots [29,30]. In addition to Pb and Cd, As is also found predominantly in the environment and is carcinogenic in nature [31]. Increases in the concentration of As can result in the absence of seed germination, stunted growth, and a decrease in the dry weight of plants [32]. Moreover, Cu is known as an essential TE required for plant metabolic functions, such as counterbalancing redox reactions. However, Cu has shown a potential risk of toxicity in plants [33]. Excess Cu can induce reactive oxygen species (ROS) in plants and affect the photosystems in the photosynthesis process. Furthermore, Cu toxicity can lead to chlorosis and reduction of the total chlorophyll content in plants [34]. Likewise, Cr can decrease the biomass of plant cells, alter the soil-microbial population, and eliminate the nutrient assimilation process [35]. Cr can cause detrimental effects on the environment by accumulating easily even at low doses.

2.2. PPCPs

ECs can be defined as any chemicals or products of chemical origin used widely by humans. They are normally present in the environment at a low concentration range (ng L^{-1} to $\mu\text{g L}^{-1}$), but can pose detrimental effects on living organisms. The accumulation of ECs in food crops has been studied under different (experimental/field) conditions [11,36,37]. The major ECs classes include PPCPs, pesticides, food preservatives, polyaromatic hydrocarbons, microplastics, and nanomaterials. In this section, we focus on the most prominent EC class found to have accumulated in vegetables, that is PPCPs. These ECs have shown significant uptake and translocation from soil to vegetable plants as documented in the literature [8,38,39]. PPCPs as mentioned above include pharmaceuticals and personal care products (PCPs). Pharmaceuticals are medicinal compounds of chemical origin used for the treatment and curing of diseases. Pharmaceutical drugs can inhibit soil microbial activity, soil respiration, seed germination, and the growth of plants [40]. They can target ion channels and may inhibit the transport of the essential enzymes and nutrients required for plant growth [12]. Li et al. [41] stated that pharmaceutical compounds can transfer into plants through contaminated soil. Pharmaceuticals include a broad range of products including antibiotics, cytostatic drugs, anti-inflammatory drugs, hormones, stimulant drugs (e.g., caffeine), β -blockers, and antiepileptic drugs. Antibiotics such as tetracycline, sulfadimidine, oxytetracycline, chloramphenicol, trimethoprim, tylosin, and erythromycin are biologically active compounds used for treating bacterial infections [10]. Anti-epileptic drugs such as carbamazepine, dilantin, and primidone, have also been widely reported in the environment due to their inordinate use. Carbamazepine is one of the most thoroughly investigated pharmaceutical drugs in plant uptake of PPCPs [42]. Furthermore, anti-inflammatory drugs including diclofenac, ibuprofen, naproxen, acetaminophen, and β -blockers such as atenolol and propranolol, have been found to have accumulated in vegetable plants [10].

Besides pharmaceuticals, PCPs include daily life and household items such as plasticizers widely used in plastic bottles, gels, shampoo, and other beautification/cosmetics products, synthetic musk including galaxolide used in fragrances, skin care products, preservatives (such as parabens), ultraviolet filters (such as oxybenzone), and antimicrobial products (such as triclosan and triclocarban) [43]. Triclosan can increase fungal diversity, decrease the soil respiration process, and reduce plant growth [44]. Triclosan and triclocarban are the most widely used PCPs that have been identified in plants and generally found in the concentration range of 10–40 mg kg^{-1} [36]. The potential uptake of PPCPs through soil and vegetable plants has paved the way for its presence in humans.

3. Sources of TEs and PPCPs in Vegetables

There are various sources of TEs and PPCPs, which can broadly be categorized into natural or anthropogenic sources. Although soil contamination by natural phenomena (e.g., weathering of rocks, volcanic eruptions, and soil erosion, etc.) is unavoidable, the increase in soil contamination by anthropogenic sources (Figure 1) can at least partially be controlled and managed. A summary of the various sources of TEs and PPCPs is discussed in this section.

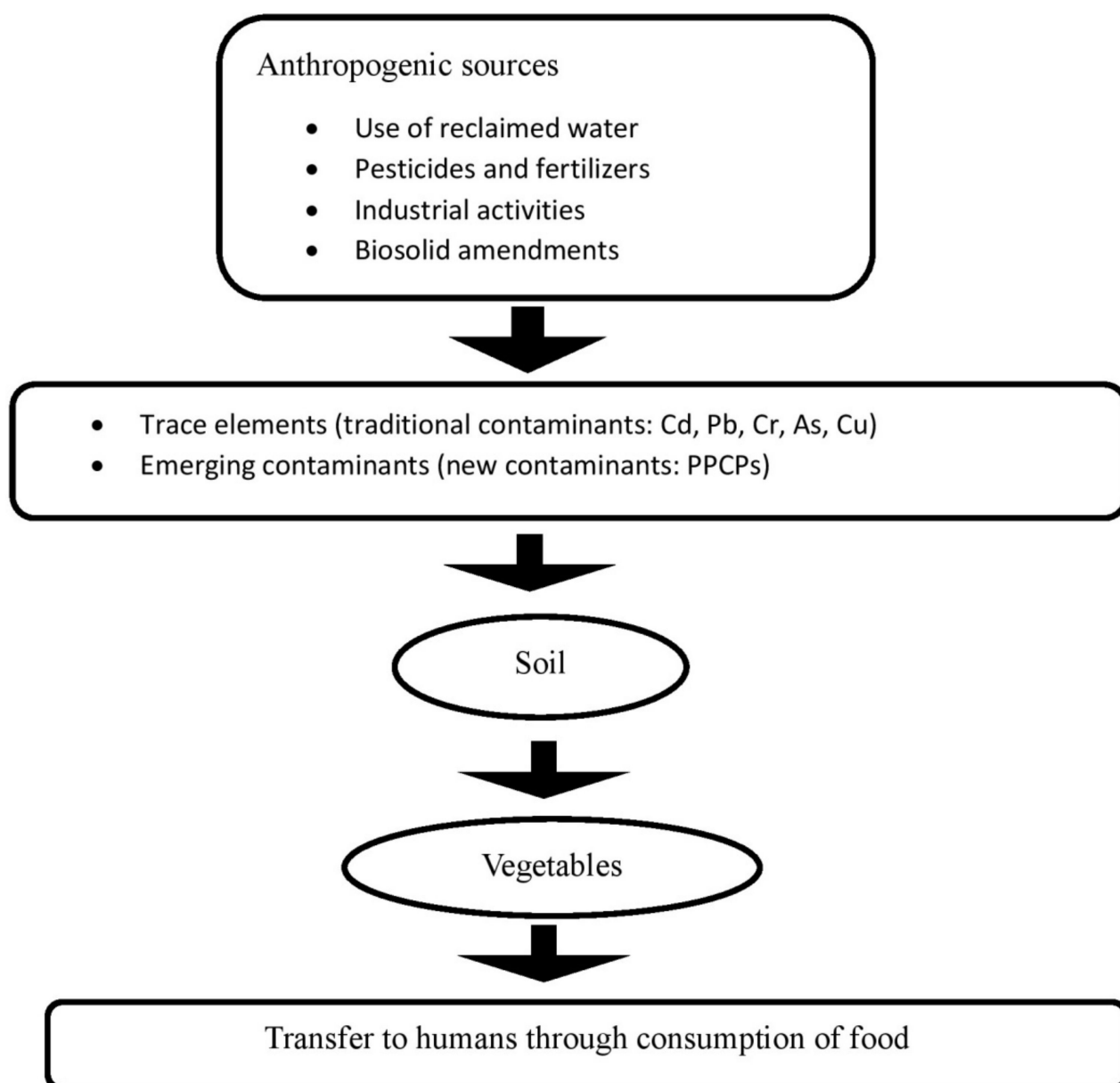


Figure 1. Anthropogenic sources of TEs and PPCPs in the soil–vegetable interface.

3.1. Industrial Activities

TEs can accumulate in the environment due to industrial activities. Industrial discharge either in the form of contaminated gases or waste effluents can contribute to TE-associated contamination [45]. In this respect, anthropogenic sources of TEs and PPCPs in vegetables reported from different studies are summarized along with their geographical area in Table 1. TE uptake in soil can occur from industrial wastes released via direct or indirect means [46]. For instance, foliar uptake of TEs was investigated in cabbage and

spinach grown near a smelter in relation to industrial activity [47]. Similarly, in a field experiment by Naser et al. [48] Cd and Pb concentrations were observed in spinach and tomatoes grown near a polluted industrial area. TE (Pb, As, Cu, and Cd) accumulation was also seen in leafy vegetables (e.g., lettuce, cabbage, spinach, and edible amaranth) grown near a mining site in China [49].

Table 1. The different types of TEs and PPCPs studied in vegetable crops.

S. No.	Class	Type	Sources	Vegetable Species	Area	References
A. TEs						
1		Cu, Cr	Wastewater irrigation	Lettuce, radish, carrot	Dubai	[5]
2		Cd, Cu, Cr, Pb	Sewage water irrigation	Tomato	India	[50]
3		As, Pb	Contaminated soil	Lettuce, carrot, tomato, radish	USA	[51]
4		Cd	Contaminated soil	Spinach, lettuce, celery, carrot	China	[52]
5		Cd, Pb, As, Cu	Mining activity	Spinach, cabbage, amaranth, lettuce	China	[49]
6		Cu, Cr, Pb	Wastewater irrigation	Onion, tomatoes, garlic, eggplant	Pakistan	[53]
7		Cd, Pb	Industrial activity	Spinach, tomato	Bangladesh	[48]
8		Cu, Pb, Cd	Agriculture activity	Spinach	India	[54]
9		Pb, Cd	Industrial activity	Cabbage, spinach	France	[47]
10		Pb	Wastewater irrigation	Cauliflower, radish, spinach	Pakistan	[55]
11		Cr, Cd, Cu, As, Pb	Agriculture activities	Cabbage, lettuce, carrot, tomato, potato	Iran	[56]
12		Cd, Pb, Cr, Cu	Wastewater irrigation	Onion, tomato	Iran	[57]
B. PPCPs						
1		Tetracycline	Wastewater irrigation	Carrot, lettuce	Ghana	[58]
2		Ibuprofen, diclofenac	Biosolid amendment	Soybean	Sweden	[59]
3		Carbamazepine, lamotrigine, caffeine	Wastewater irrigation	Carrot	Israel	[60]
4		Carbamazepine	Biosolid amendment	Cabbage	USA	[61]
5		Sulfamethazine	Contaminated soil	Cabbage	China	[62]
6		Chloramphenicol	Contaminated soil	Eggplant	China	[63]
7		Triclosan	Biosolid amendment	Lettuce	USA	[64]
8		Triclocarban	Biosolid amendment	Carrot	USA	[65]
9		Triclosan, galaxolide	Biosolid amendment	Carrots	Germany	[66]
10		Bisphenol-A	Wastewater irrigation	Lettuce	USA	[67]
11		Triclosan, triclocarban	Wastewater irrigation	Carrot, celery, lettuce, spinach, tomato	USA	[68]

3.2. Agricultural Practices

Agricultural practices include the use of pesticides, herbicides, or inorganic/organic fertilizers which can contribute to TE contamination [13]. Fertilizers are added in soil to improve growth of plants as well as provide N, P, and K in plants [69]. Trace amounts of Pb, Cd, As, Cr, and other chemicals are already present during manufacturing of fertilizers. Their indiscriminate use in plants can easily translocate TEs into the soil. Cd is one such contaminant of concern due to its high mobility and translocation capability in soil and plants [13]. In the same manner, pesticide application to control pests and increase crop yield can also induce TE accumulation [70]. Once pesticides are sprayed on the soil, they can translocate into the edible portions of vegetable crops through the roots [71]. For example, excessive usage of DELVAP 100EC, a pesticide residue, increased TE concentrations in spinach leaves [72].

3.3. Wastewater Irrigation

Reclaimed/recycled water such as sewage or any wastewater is widely used as an alternative to typical irrigation water in order to conserve water resources [73]. However, this alternative can induce or increase both TE and PPCP toxicity. The accumulation of TEs in soils and vegetable crops cultivated with various sources of reclaimed/recycled water irrigation has been documented in many studies [4,6,74]. For example, elevated levels of TEs (i.e., Cd, Pb, Cu, and Cr) were found in tomatoes cultivated with treated sewage water in a field study [50]. Another study detected TEs, such as Cr, Cu, and Pb in vegetables (onion, garlic, eggplant, and tomato) irrigated with wastewater [53].

Reclaimed/recycled water not only induces the contamination of TEs in vegetable crops, but also increases the PPCP concentrations via agricultural irrigation. Incremental contamination in reclaimed/recycled water applied to crops has become one of the prime sources for their uptake in the soil–plant system [12]. Recycled water from treated plants releases micro-contaminants into the soil–plant system that penetrates the tissues of edible crops [60,75]. A study reported PPCPs (carbamazepine and diclofenac) in the leaves of iceberg lettuces which was irrigated with treated wastewater [76]. Similarly, the potential uptake of PPCPs was found in carrots due to wastewater irrigation in a study [60]. Likewise, Azanu et al. [58] explored the uptake of tetracycline by carrot and lettuce plants with the introduction of treated wastewater

3.4. Biosolid Amendments

Biosolids are solid organic matters produced from sewage treatment plants. Sludge obtained from sewage treatment plants is often used as fertilizer and manure in agriculture, which has become a source of TEs and PPCPs being transferred into the soil and vegetables [59]. Studies have shown that biosolids have the potential to accumulate TEs and PPCPs in the soil–vegetable system leading to health risks [36,61,77]. For instance, amendments of soybean plants with biosolids led to the uptake of diclofenac and ibuprofen in a study by Cortes et al. [59]. Similarly, PPCPs (i.e., carbamazepine, diphenhydramine, and triclocarbon) were reported in vegetables (e.g., tomato, lettuce, and radish plants) treated with biosolids [78]. Another study found the accumulation of 15 PPCPs (i.e., carbamazepine, sulfamethoxazole, acetaminophen, estrone, sulfadiazine, caffeine, lincomycin lamotrigine, carbadox, triclosan, 17 β -estradiol, trimethoprim, monensin, oxytetracycline, and tylosin) in the soil–radish system due to biosolid amendments [20]. Biosolid amendments have also been reported to accumulate TEs such as Pb, Cd, Cu, As, and Cr in the soil, which can transfer into vegetable plants easily [69]. For instance, the health risk assessment was analyzed for Cu and Cd in vegetables amended with biosolids [79]. The authors concluded that there was a potential risk of exposure due to biosolid amendment in the soil. In another study, the uptake of TEs (i.e., Cd, Pb, and Cu) was found in lettuce grown in soil amended with biosolids [80].

4. Factors Influencing the Uptake of Contaminants

4.1. Soil Factors

The interaction between the soil and plants determines the uptake of TEs and PPCPs. Soil plays an essential role in plant growth and development; however, it can also accumulate or regulate the toxic contaminants. Various physio–chemical and biological properties govern the functioning of the soil, such as pH, soil organic content, soil moisture, CEC, redox potential, enzymes, temperature, texture, mobility, and aging [8,13,81]. Soil pH is considered important in determining the solubility/mobility of the soil contaminants. An inverse relationship has been observed between soil pH and contaminant solubility [82]. The solubility of TEs and PPCPs decreases at high pH levels and increases at low pH levels due to surface charge variation and adsorption of solutes with charged soil components [82]. However, TEs often tend to show differences in uptake with respect to each other. For example, in the case of As, its solubility increases with increasing pH levels [83]. Due to ionic interaction present between the elements, another factor influencing the TE and PPCP movement in the soil is the redox potential (Eh) [84]. The Eh of soil decreases with the addition of organic fertilizers. The bioavailability also decreased for Cd and increased for As in soil with a decreasing Eh with organic amendments [83]. CEC is another significant factor present in the uptake of contaminants and depends purely on the soil content. For instance, soil with a high clay content increases the CEC of soil and decreases the availability of contaminants [13]. In contrast, sand decreases the CEC and increases the mobility of contaminants in the soil. Organic amendments like biochar or peat also increase the CEC of soil [85].

Furthermore, soil organic matter also influences the uptake of TEs and PPCPs. The organic matter of soil contains humic substances, such as humic and fulvic acids. Humic acid is insoluble at high pH levels, while fulvic acid is soluble in all pH conditions. As such, fulvic acid shows an anionic nature with active functional groups. Therefore, due to the presence of these humic substances, soil shows anionic properties and can bind strongly with cations through the ion-exchange reaction [86]. The main mechanisms involved in the retention of TEs and PPCPs in soil are adsorption and ion-exchange reactions. Thus, humic substances are determinants for the retention of TEs and PPCPs in soil [8,13].

Soil temperature is also a critical factor in determining the uptake of TEs and PPCPs. High temperatures enhance the uptake process for the contaminants. For instance, it was reported in a study that Cd translocation from soil to plants increased with an increase in soil temperature [87]. In addition, an increase in soil organic content may also lead to an increase in soil temperature which enhances the bioavailability of these contaminants in the soil. Soil moisture content also influences the availability of different soil contaminants. For instance, in a study by Stafford et al. [88], Cd solubility was affected by a regular increase in soil moisture and remained unaffected by irregular changes in the duration of soil saturation.

4.2. Plant Factors

In addition to soil factors, plant mediated pathways also influence the uptake of contaminants. The uptake of TEs and PPCPs mainly occurs with the help of the roots and leaves of the plants [89]. Root uptake is affected by the hydrophobicity and cell wall permeability [86] since the cell wall restricts the movement of the pollutant molecules based on size. The molecular weight plays an important role in determining the contaminant uptake through the roots. Pollutants with low molecular weights can easily move through the cell wall pores while high molecular weight compounds have low uptake rates in plants [86]. Moreover, active and passive transport are the major pathways helping in the uptake of contaminants into the plants. Active transport initiates the movement of contaminants with the help of carrier proteins through the plasma membrane. In addition, passive transport occurs along the intercellular spaces of the plant root cells with the help of diffusion [90]. Soil contaminants enter the aerial parts of the plant through the xylem with the help of transpiration and accumulate in the vegetative tissue via the phloem [91]. All these processes are regulated with the help of binding compounds and transport proteins, such as yellow stripe (YS), heavy metal transporting ATPases (HMAs), phytochelatins, ferroportin (FPN), copper transporter (COPT), and cation exchanger (CAX), for the uptake [13].

In contrast to root uptake from soil, foliar uptake of TEs and PPCPs depends on the morphology of plants, such as leaf area, leaf inclination angle, size of the stomata, and cuticular movements [92]. The stomatal system serves as the major regulator in the infiltration process for contaminants due to the presence of apertures. The opening and closing of the stomatal aperture depends on external and internal factors, such as light, humidity, temperature, partial pressure of CO₂ in the intracellular space, and ionic condition of the plants [86]. Due to these factors, aperture diameters are not fixed and can be modified with the variation of the aforementioned factors. It has been observed that vegetable plant uptake of contaminants is inconsistent due to the differences in anatomical and morphological characteristics of the plants [81]. It has also been reported that leafy vegetables tend to accumulate more TEs and PPCPs than other varieties of vegetables due to their greater transpiration rate [13].

5. Accumulation Concentration of TEs and PPCPs in Vegetables

5.1. TEs

Many studies have documented the accumulation of TEs in vegetable crops [13,25,93]. The concentration level of TEs in vegetables reported by various researchers is summarized in Table 2 and some are discussed below. TE (i.e., Cu and Cr) accumulation was reported

in lettuce grown near a waste-dumping and uncontrolled burning area in the Campania lands in Italy [94]. The study reported accumulation at 0.898 and 0.085 mg kg⁻¹ for Cu and Cr, respectively. In another study conducted in China, the uptake of TEs, such as Pb, Cd, As, and Cr was investigated in the underground portion of Chinese cabbages grown in pots under greenhouse conditions [95]. These authors found average concentrations for Pb, Cd, As, and Cr of 1.62, 0.67, 1.36, and 0.73 mg kg⁻¹, respectively, in the cabbages. These values were above the permissible limits for TEs given by various regulatory agencies (Table 3). Similarly, the accumulation capacities of Cu, Cd, and Cr were studied in onions consumed and distributed in the local markets of Ghana [96]. Among the studied TEs, the concentrations ranged from 0.03–0.06 mg kg⁻¹, 0.13–0.52 mg kg⁻¹, and 0.01–0.07 mg kg⁻¹ for Cd, Cr, and Cu, respectively, in the onions. Furthermore, the levels of TEs were analyzed in dietary surveys for vegetables cultivated near a mining complex in Armenia [97]. This study determined the uptake levels for As, Cu, and Pb were at 0.03, 12.01, and 0.54 mg kg⁻¹, respectively, in potatoes. However, in carrots, the uptake levels for Cu and Pb were found to be 5.78 and 0.54 mg kg⁻¹, respectively. Another recent study reported TE (Pb, Cr, Cu, and Cd) accumulation in vegetables (cabbage, tomato, pepper, carrot, and lettuce) grown in soils of Ethiopia. The results found high mean concentrations of Cd in all vegetables within the range of 0.20–0.38 mg kg⁻¹ [98].

Table 2. Concentration of TEs (mg kg⁻¹) studied in different vegetables.

S. No.	Vegetables	Botanical Name	Cd	Pb	As	Cu	Cr	Range	References
1	Bottle gourd	<i>Lagenaria siceraria</i>	0.022	0.34	0.036	7.48	0.47	0.022–7.48	[99]
2	Potato	<i>Solanum tuberosum</i>	-	0.54	0.03	12.01	-	0.54–12.01	[97]
3	Carrot	<i>Daucus carota</i>	-	0.69	-	5.78	-	0.69–5.78	[97]
4	Cabbage	<i>Brassica oleracea</i>	<0.50	-	-	0.16	<0.50	0.16–<0.50	[100]
5	Lettuce	<i>Lactuca sativa</i>	-	-	-	0.898	0.085	0.085–0.898	[94]
6	Coriander	<i>Coriandrum sativum</i>	<0.07	1	<0.19	16.5	20.1	<0.07–20.1	[101]
7	Eggplant	<i>Solanum melongena</i>	0.07	1.2	0.33	-	1.8	0.07–1.8	[102]
8	Radish	<i>Raphanus sativus</i>	0.1	0.63	-	0.66	<0.05	<0.05–0.66	[103]
9	Onion	<i>Allium cepa</i>	0.36	1.39	2.95	-	1.01	0.36–2.95	[104]
10	Spinach	<i>Spinacia oleracea</i>	2.67	2.01	-	1.79	0.82	0.82–2.67	[105]
11	Cabbage	<i>Brassica oleracea</i>	-	0.07	0.003	0.99	0.16	0.003–0.99	[106]
12	Eggplant	<i>Solanum melongena</i>	0.25	0.7	0.78	29	1	0.25–29	[107]
13	Onion	<i>Allium cepa</i>	0.05	-	-	0.06	0.32	0.05–0.32	[96]
14	Lettuce	<i>Lactuca sativa</i>	0.98	1.1	1.16	-	0.6	0.6–1.16	[108]
15	Chinese Cabbage	<i>Brassica rapa pekinensis</i>	0.67	1.62	1.36	-	0.73	0.67–1.62	[95]
16	Spinach	<i>Spinacia oleracea</i>	0.022	0.016	0.004	0.861	-	0.004–0.861	[93]
17	Tomato	<i>Solanum lycopersicum</i>	0.013	0.084	0.009	-	-	0.009–0.084	[93]
18	Coriander	<i>Coriandrum sativum</i>	-	0.114	0.037	-	0.142	0.114–0.142	[25]
19	Cabbage	<i>Brassica oleracea</i>	0.017	0.030	0.012	-	0.080	0.012–0.080	[25]
20	Radish	<i>Raphanus sativus</i>	0.087	0.980	0.150	-	0.130	0.130–0.980	[25]

Table 3. Permissible limits by various organizations for TEs in vegetables (mg kg⁻¹).

S. No.	TEs	WHO	EU	Indian Standard
1	As	0.2	0.2	1.1
2	Pb	0.3	0.43	2.5
3	Cd	0.2	0.2	1.5
4	Cu	40	20	30
5	Cr	20	1	20

5.2. PPCPs

Accumulation of PPCPs in plants is often expressed using the bio-concentration factor (BCF), root concentration factor (RCF), and translocation factor (TF). BCF is defined as the ratio of the concentration in plant tissues to that of the spike concentration given in the media [36]. RCF is evaluated precisely for PPCP accumulation in roots and is elucidated as the ratio of the concentration in the roots to that of the spiked concentration. Likewise, TF is expressed as the ratio of the concentration in the aerial plant parts to that in the roots [12]. A number of studies have been performed regarding the accumulation of PPCPs

under greenhouse conditions (i.e., hydroponic and pot experiments) [75,109–111] and few under field conditions [11,63,77]. For example, in a field study, the accumulation of PPCPs such as trimethoprim, sulfamethoxazole, and diclofenac, was reported at 0.572, 0.406, and 3.863 $\mu\text{g kg}^{-1}$ respectively, in tomatoes irrigated with wastewater [11]. In another study PPCP uptake in eggplants grown in wastewater was investigated [63]. The study reported the occurrence of triclosan, chloramphenicol, ibuprofen, and trimethoprim in the range of 0.02–28.1 $\mu\text{g kg}^{-1}$, respectively, in the eggplants. PPCP (i.e., carbamazepine, caffeine, and ibuprofen, triclosan, and bisphenol-A) accumulation was also reported in lettuces with varying spike concentrations (0–40 $\mu\text{g L}^{-1}$) [112]. The authors reported a concentration in lettuces at the range of 0.93–2054 ng g^{-1} . However, of all the studied PPCPs, carbamazepine showed the highest accumulation and maximum translocation in the lettuce leaves. The concentration of carbamazepine at different spiked concentrations (i.e., 0–40 $\mu\text{g L}^{-1}$) was in the range of 233–2054 ng g^{-1} in the leaves of lettuces. In another study, the uptake of PPCPs (i.e., caffeine, carbamazepine, meprobamate, and primidone) was investigated in celery plants irrigated with wastewater [68]. These authors reported the concentrations of caffeine, carbamazepine, meprobamate, and primidone in celery plants at 0.17, 0.50, 0.20, and 0.45 ng g^{-1} , respectively, when determined on a dry weight basis. The concentrations of various PPCPs reported by several researchers in the edible parts of vegetables are shown in Table 4.

Table 4. Concentration of PPCPs ($\text{ng g}^{-1}/\mu\text{g kg}^{-1}$) studied in different vegetables.

S. No.	Vegetable	Botanical Name	PPCPs	References
1	Lettuce	<i>Lactuca sativa</i>	Carbamazepine-233 ng g^{-1} Ibuprofen-0.93 ng g^{-1} Triclosan-13 ng g^{-1} Bisphenol-A-33 ng g^{-1} Caffeine-32 ng g^{-1}	[112]
2	Tomato	<i>Solanum lycopersicum</i>	Diclofenac-3.863 $\mu\text{g kg}^{-1}$ Trimethoprim-0.572 $\mu\text{g kg}^{-1}$ Sulfamethoxazole-0.406 $\mu\text{g kg}^{-1}$	[11]
3	Cabbage	<i>Brassica oleracea</i>	Naproxen-38 ng g^{-1} Gemfibrozil-75 ng g^{-1} Atenolol-55 ng g^{-1}	[113]
4	Eggplant	<i>Solanum melongena</i>	Caffeine-125 ng g^{-1} Gemfibrozil-34 ng g^{-1}	[113]
5	Tomato	<i>Solanum lycopersicum</i>	Atenolol-3.95 ng g^{-1} Meprobamate-0.20 ng g^{-1} Minocycline-13.83 ng g^{-1}	[77]
6	Celery	<i>Apium graveolens</i>	Carbamazepine-0.50 ng g^{-1} Caffeine-0.17 ng g^{-1} Primidone-0.45 ng g^{-1} Diclofenac-9.05 ng g^{-1}	[68]
7	Lettuce	<i>Lactuca sativa</i>	Bisphenol-A-0.36 ng g^{-1} Naproxen-2.81 ng g^{-1}	[110]
8	Cabbage	<i>Brassica oleracea</i>	Carbamazepine-19.34 ng g^{-1} Salbutamol-11.34 ng g^{-1} Trimethoprim-11.42 ng g^{-1}	[114]
9	Tomato	<i>Solanum lycopersicum</i>	Sulfamethoxazole-20.10 ng g^{-1} Bisphenol-F-50.5 $\mu\text{g kg}^{-1}$ Carbamazepine-0.12 $\mu\text{g kg}^{-1}$	[3]

6. Effect of TEs and PPCPs on the Growth of Vegetables

6.1. TEs

TEs can be accumulated in almost all vegetable parts including the root, stem, leaves, and tubers. It is reported that leaves tend to absorb a maximum amount of TEs followed by the root and stem [13]. TE toxicity can inhibit enzymatic activity and induce oxidative stress in the vegetables [115]. Pb and Cd can interfere and halt the transportation and absorption of other essential elements in vegetables, which can disrupt the photosynthesis process, electron transport chain, respiration rate, and growth in vegetables [116]. TE toxicity can also result in the reduction in leaf area, biomass, dry weight, shoot growth, and yield in

vegetables [117]. Other prominent effects on the growth of vegetables includes an increase in oxidative and peroxidase enzymes along with a decrease in anti-oxidant activities [13]. TEs are independent and show different effects on different vegetables. For instance, it was found that Cd toxicity can alter root growth in potato and decrease protein content in carrot leaves [118] and can reduce fruit production in tomatoes [119]. Similarly, As can cause a reduction in photosynthetic pigments as reported for carrots, lettuces, and spinach [120]. Cr toxicity can impair seed germination and inhibit plant growth, can cause chlorosis in young leaves as investigated in onions, and induce nutrient imbalance and injury as studied in tomatoes [121]. In a study by Hamid et al. [122], it was found that Pb toxicity can increase the formation of the harmful chemical compound lead acetate in beans, and reduce chlorophyll content in peas. Similarly, TEs such as Cu may alter root development and inhibit the micronutrients activity [13].

6.2. PPCPs

The effect of PPCPs on vegetables depends on the specific class type of PPCPs, the concentration, and the vegetable species exposed [39]. The prominent effect on vegetables due to PPCP toxicity includes decreases in reproduction, reduction in root length, chlorosis, decreased biomass, increases in oxidative stress, and inhibition of growth [39,123]. For example, carbamazepine toxicity can result in burnt edges, decreased photosynthesis pigments, and white spots in vegetable plants [124]. The effect of industrial pharmaceuticals on the growth and germination of spinach through wastewater irrigation was studied by Islam et al. [125]. The results showed 92% germination rate in control conditions in contrast to PPCP contamination at only a 21% germination. Similarly, the roots and shoot length were significantly decreased in the PPCP-contaminated spinach. The authors concluded that PPCP contamination has deleterious effects on vegetables, which lowers the yield, growth, and biomass. Therefore, to avoid PPCP contamination, the proper treatment of effluent is critical before using it for irrigation. The effect of ten antibiotics was studied in carrots and lettuces to investigate their influence on seed germination and root and shoot length [126]. The results showed a negative impact on seed germination with a decrease in root vegetable growth. Goldstein et al. [127], stated that PPCPs such as carbamazepine have the maximum tendency to accumulate more in the leaves than in other parts of plants.

7. Health Risk through Consumption of Contaminated Vegetables

7.1. TEs

Consumption of contaminated vegetables can affect human health. The list of adverse effects on human health due to consumption of TE and PPCP-contaminated vegetables is presented in Table 5. Common symptoms due to TE toxicity on human health include mental retardation in children, malnutrition, central nervous disorders, decreased intra-uterine growth, dementia, insomnia, depression, liver and kidney disorders, weak immune systems, instability, decreased vision, gastrointestinal disorders, cancer, and death [13]. Regular exposure to Cd can lead to bronchiolitis, alveolitis, emphysema, and other respiratory diseases [128]. Certain neurological and mental illnesses can be induced especially in children due to Pb toxicity [129]. Moreover, Pb and Cd can readily accumulate in bone matrices and tissues and induce fractures and bone deformities. It can also cause malfunctioning of the lungs and liver and can affect other metabolic functions of the human body [49,130]. The toxicity range depends on the dietary intake of the contaminated vegetables and is assessed using health risk parameters. The various indices and parameters used in the assessment of the health risks posed by TEs and PPCPs is shown in Table 6. For TE-contaminated vegetables, the health risk assessment for humans is determined by parameters such as the daily intake of metals (DIM), hazard quotient (HQ), target hazard quotient (THQ), and health risk index (HRI) [13]. THQ measures the health risk posed by TEs through vegetable consumption. HRI represents the toxicity index, where a value <1 shows it is safe for the population and a value >1 shows detrimental effects on the population. For instance, a high HRI value was reported in a study on vegetables grown

near Pb mines upon regular consumption of leafy vegetables. The risk assessment study revealed an HRI value >1 indicating a risk to consumers for several health problems, such as Alzheimer's disease, due to excessive Pb intake through vegetables [129].

Table 5. Risk effects of TEs and PPCPs on human health.

S. No.	Contaminants	Health Risk Effect	References
A			
TEs			
1	Cd	Causes various types of cancer, prostate and breast cancer common, liver and kidneys most sensitive to Cd exposure, damage to hemopoietic system, rheumatoid arthritis, osteoarthritis, osteoporosis, renal and hepatic dysfunction, and pulmonary edema. Excess Cd can cause damage to lungs. It can induce neurodegenerative disease, affect male reproductive system, female menstrual cycle, impair reproductive hormones, fertility, pose a threat to the life of pregnant women and the newborn baby, and cause death.	[130–132]
2	As	Causes a variety of complications in the body's organ system, such as the nervous, respiratory, cardiovascular, and integumentary systems, skin lesions, bone marrow depression, and encephalopathy. As is highly carcinogenic in nature causing various types of cancer including skin, bladder, lung, kidney, and liver cancer. Neurological disorders include intellectual function and memory loss, arsenicosis, diabetes, hypertension, hyperkeratosis, and melanosis.	[133,134]
3	Pb	Pb affects the central nervous system, cardiovascular system, and kidneys; pregnancy complications can occur, stopping fetal growth in the early stages of pregnancy, miscarriage in females, and male infertility; children have a high absorption capacity for Pb, it can pass through placenta causing birth defects, and can reduce maternal thyroid hormones. Pb can induce anemia, hypertension, renal dysfunction, neurotoxicity, heme synthesis, nephrotoxicity, and encephalopathy and can induce oxidative stress in the liver; excess Pb causes chronic respiratory and dermatogenic problems. It can act as substitute for Zn required in heme synthesis. It can cause various types of cancer due to its carcinogenic nature, leading to death	[129,135,136]
4	Cu	Affects liver and kidney failure through chronic or long-term exposure, diarrhea, headache, and mucosal irritation. High Cu concentrations can cause gastrointestinal disorders like jaundice, stomach pain, hematemesis, and vomiting. Other symptoms include hepatic necrosis, rhabdomyolysis, intravascular hemolysis, encephalopathy, depression, irritation, and mortality. Cu can also induce DNA damage due to oxidative stress	[137–139]
5	Cr	Chromosomal abnormalities, DNA strand breakage, risk of abortion or miscarriage, respiratory tract irritants can cause pulmonary sensitization, and the chronic inhalation of Cr (VI) causes risk of ulcers, lung or nasal cancer. Cr toxicity can induce stomach, kidney, bones, lungs, and other gastrointestinal disorders. It can impair broad spectrum alteration in DNA causing neoplasia	[13,140]
B	PPCPs	PPCPs have been detected in environmental entities due to their application as prophylactics, and growth promoters in food. Antibiotics distribution in vegetables have been found in the order of leaf > stem > root. Daily exposure to PPCPs can cause antibiotic resistance in humans which can increase the risk of death. Parabens suppress estrogen activity and induce an immediate hypersensitivity reaction. Many laboratory experiments have suggested a low risk of PPCPs exposure on human health. There still needs to be more field scale data under the current system of agricultural production for the analysis of PPCP toxicity on humans through the ingestion of contaminated vegetables.	[10,141]

In addition, high levels of Cu exposure can cause anemia, and liver and kidney damage [137]. It is also associated with a genetic disorder called Wilson disease when it accumulates in the liver [142]. Cr in the form of Cr⁺⁶ is considered to be detrimental for human health due to its toxicity and carcinogenic nature [143]. Furthermore, Smith et al. [144] stated that the consumption of As-contaminated vegetables even at low concentrations can lead to irregularity in heartbeat, nausea, vomiting, and low red-blood cells (RBCs), and white-blood cells (WBCs) counts. In addition, high concentrations of As in vegetables can cause diabetes, cardiovascular disease, an increase in blood pressure, and neurological and pulmonary disorders as well as cancers when consumed regularly.

7.2. PPCPs

Imperceptible amounts of PPCPs in dietary intakes can cause allergies, especially in children. The health risks associated with the consumption of PPCP-contaminated vegetables are shown in Table 5. Long-term consumption of PPCP-contaminated vegetables, especially antibiotics, can lead to the development of resistance against natural human

antibiotic activity, which can cause illnesses that are difficult to cure and can lead to death [109]. Likewise, Stuart et al. [43] reported in a review regarding the risk assessment of ECs that paraben toxicity can decrease estrogen activity and increase hypersensitivity reactions.

Table 6. Toxicity assessment parameters for TEs and PPCPs.

S.No.	Parameters	Formula
A TEs		
1	Daily intake of metal (DIM)	$DIM = A * C * D / B_W$ A = TEs concentration in plants ($mg\ kg^{-1}$) C = conversion factor D = daily intake of vegetable ($kg\ day^{-1}$) B _W = average human body weight (kg)
2	Hazard quotient (HQ)	$HQ = (\text{daily vegetable intake}) * (\text{metal concentration in vegetable}) / R_{FD}$ (oral reference dose in $mg\ kg$) * body mass
3	Health risk index (HRI)	$HRI = DIM / R_{FD}$ $THQ = 10^{-3} (E_F * E_D * F_{IR} * C / R_{FD} * W_{AB} * T_A)$ E _F = exposure frequency (365 days year ⁻¹) E _D = exposure duration (70 years)
4	Target hazard quotient (THQ)	F _{IR} = food ingestion rate ($g\ person^{-1}\ day^{-1}$) C = metal concentration in food ($\mu g\ g^{-1}$) R _{FD} = oral reference dose ($mg\ kg^{-1}\ day^{-1}$) W _{AB} = average body weight (Kg) T _A = average exposure time for non-carcinogens
B PPCPs		
1	Estimated daily intake (EDI)	$EDI = \text{Daily intake rate } (g\ person^{-1}\ day^{-1}) * \text{PPCPs in vegetables } (ng\ g^{-1}) / \text{person average weight } (kg\ person^{-1})$
2	Risk quotient (RQ)	$RQ = EDI (ng\ Kg^{-1}\ day^{-1}) / ADI (ng\ Kg^{-1}\ day^{-1})$
3	Health hazard index (HHI)	$HI = \sum_{i=1}^n RQ_i$ $HE = C * D * W * T$ C-concentration of PPCPs in vegetables (ng/kg)
4	Human exposure (HE)	D-average daily consumption of vegetables (Kg/day) W-human body weight (Kg) T-exposure time (day)

However, many laboratory studies have suggested a low risk of PPCPs on human health through vegetable consumption [7,10]. The annual human exposure value for PPCPs (i.e., carbamazepine, diclofenac, triclosan, and parabens) is acceptable in the range of 20–200 mg [10]. For instance, in a greenhouse experiment, a low annual exposure of PPCPs was reported for spinach in the range of 0.04 to $3.5 \times 10^2\ \mu g$ and for lettuce in the range of 0.08 to $1.5 \times 10^2\ \mu g$ for an average 70 kg individual [111]. These estimates in the study were much lower than the acceptable range (i.e., 20–200 mg) and showed a low risk of exposure through vegetable consumption. Similarly, triclosan exposure was found to cause minimal risk to human health in a study evaluating the health risk assessment [145]. To date, except for the laboratory-based calculations, there is not enough data under realistic field conditions to make conclusions for human safety regarding exposure to PPCPs via contaminated vegetable consumption [146]. Furthermore, the risks associated with PPCP-contaminated vegetables are measured in terms of the risk quotient (RQ) and health hazard index (HHI) (refer to Table 6). RQ is defined as the ratio of the estimated daily intake (EDI) to the accepted daily intake (ADI) and HHI is calculated by the summation of the RQ of each of the PPCPs. When the RQ and HHI value is <0.01, PPCP-exposure risk to humans is negligible, while when the RQ and HHI value is >0.01 it shows the possibility of risk; when this value is >0.05 it shows a high risk for humans [10]. For example, in a study on the accumulation of pharmaceuticals in peanut kernels, their daily intake was explored with respect to their potential risks to human health [147]. The study reported a high risk

of exposure with $HHI > 0.05$ for enrofloxacin. However, for norfloxacin, it showed the possibility of a health risk with an $HHI > 0.01$. To conclude, more studies are required in the near future for a better understanding of the health risk imposed by PPCP-contaminated vegetables.

8. Mitigation Strategies of TEs and PPCPs through Biochar

Soil clean-up is a challenging task due to technical and financial difficulties. Various efforts have been employed for the remediation of soil to reduce the impact of pollutants from harming human health and the environment. These include physical remediation (isolation, replacement, vitrification, and electrokinetic approaches), chemical remediation (such as ion exchange, immobilization techniques, photocatalysis, and chemical precipitation), and biological remediation (microorganism-based remediation or plant-based (called phytoremediation) approaches). However, due to their respective advantages and disadvantages, these methods are not always effective [148]. Recent approaches for mitigation have been shifted to the application of amendments to the soil due to the feasibility and productive results: a low cost and effective timeframe. Biochar is such a soil amendment and has been used widely in recent years [149]. Biochar is defined as solid carbonized material which is eco-friendly and cost effective [150]. It is porous in nature and obtained from the anaerobic pyrolysis of biomass such as plant, animal, and sludge waste at a temperature in the range of 300–1000 °C [151]. Biochar has been found to be effective in the remediation of soil from both TEs and PPCPs and has gained interest recently [152]. In addition, biochar has also proved to be beneficial in improving soil nutrient quality, and plant growth, yield, and productivity [151].

8.1. TEs

Biochar amendment for TE remediation has been used widely in recent years and is estimated to be of prime importance in near future for sustainable agriculture produce [153]. Various techniques used in the characterization and analysis of biochar such as X-ray diffraction (XRD), scanning electron microscopy (SEM), and Fourier transform infrared spectroscopy (FTIR), etc., have shown high sorption capacity for TEs [154]. Application of biochar for mitigating TEs in the soil–vegetable interface has been found in various studies [122,123,124,125] [121–124]. For instance, in a study by Hmid et al. [155], Pb uptake was decreased in soil and beans due to biochar application. Moreover, it was also found that TE uptake decreased with increasing doses of biochar. Likewise, in another study it was found that the application of rice husk-derived biochar decreased the TE (Pb, Cd, Cu) concentration in lettuces [156]. In another study, the concentration of TEs (Cd, As, Cu) decreased in tomatoes after the application of biochar [157]. Important findings on biochar remediation for the TE-contaminated soil–vegetable interface is given in Table 7. Biochar contains organic functional groups, a high pH, and an effective surface area and pore structure which makes it a good adsorbent for TEs [158]. Moreover, a high pH leads to low TE mobility in soil leading to a decrease in TEs uptake in the soil and vegetables [158]. A high pH may also alter the redox state of TEs in the soil resulting in their immobilization [153]. The mechanism involved in the adsorption of TEs includes surface sorption, precipitation, complexation, ion exchange, and electrostatic interaction [154]. It was also found that the remediation of TEs by biochar depends on the type of biomass used, pyrolysis temperature, soil organic content, application of different doses, and the TEs [159]. However, more research is still required before coming to any conclusion.

Table 7. The impact of biochar amendment on vegetables contaminated with TEs and PPCPs.

S.No.	Contaminants	Raw Material	Vegetables	Average Reduction (%)	Conclusion of the Study	References
A TEs						
1	Cd	Pigeon pea	Spinach	25.50–35.06%	Biochar application increased the dry matter yield of spinach leaves and roots. It also decreased Cd mobility in both the spinach and soil.	[160]
2	Cu, Pb, Cd, Cr	Wheat straw	Lettuce, spinach, radish, parsley	51–57%	Biochar addition at a higher dose confirmed a noticeable reduction in TEs availability. However, the response of biochar varied with different TEs and vegetable species.	[158]
3	Cd	Wheat straw	Pepper, cabbage and Eggplant	11.5–15.4%	Biomass of vegetables increased and the Cd concentration decreased.	[19]
4	Cr, Cu, Pb	Hardwood	Spinach	50–75%	Hardwood biochar decreased the concentration and bioaccumulation of TEs compared to controls.	[161]
5	Cd	Bamboo and Rice straw	Chinese cabbage	Rice straw: 17–35.4% Bamboo: 12–48.3%	Cd accumulation was reduced. High concentrations of applied biochar increased nutrient uptake in vegetables.	[162]
6	Pb, Cd	Chicken manure and green waste derived	Indian mustard	Chicken manure: 74.7–96.3% Green waste: 67.2–81.6%	Biochar reduced TEs uptake and improved yield of vegetable plants.	[163]
7	Cd	Rice straw	Lettuce	-	Addition of biochar reduced the Cd uptake in shoots in slightly contaminated soil, however in heavily contaminated soil no effect on plants was found.	[164]
8	Cd Cu Pb	Rice straw	Cabbage	-	The uptake concentration of TE decreased significantly and yield increased by 34–76%.	[165]
9	As	Orchard Prune residues	Tomato	68%	A significant reduction of As in tomatoes.	[166]
B PPCPs						
1	15 PPCPs	Forest pine wood	Radish	33.3–83.0%	The uptake of 11 PPCPs out of 15 was reduced in radishes after biochar addition.	[20]
2	Triclosan	Walnut shell	Carrot	67%	Biochar addition decreased the uptake rate in vegetables, however no significant increase in biomass was reported.	[167]
3	Sulfamethoxazole	Cinnamon wood	Water spinach	30–60% in root and 61–95% in shoot	A reduced uptake was reported in plants with a biochar addition.	[168]
4	Sulphamethazine	Burcucumber plant	Lettuce	86%	Biochar application enhanced adsorption of sulphamethazine and reduced uptake in lettuces.	[169]

8.2. PPCPs

Biochar has shown good to excellent results in removing PPCP contamination due to the presence of organic and aromatic functional groups found on its surface [170]. Biochar mitigates the PPCPs by altering the pH, electrical conductivity, and total organic content of the soil which helps in its immobilization [171]. The mechanism involved for PPCP adsorption is little different from that of TE mechanisms. These include pore filling, partitioning, hydrophobic interaction, electrostatic interaction, and electron donor and acceptor interaction [154]. So far few studies regarding the amendment from biochar on the PPCP-contaminated soils–vegetable system are available. For example, biochar application on sulphamethazine-contaminated soil decreased the uptake by 86% as well as improved the lettuce yield and biomass [169]. In another study, the PPCPs such as sulfadiazine and sulfamethazine were significantly reduced with the amendment from wheat straw biochar [172]. Similarly, Williams et al. [173] found a reduction in the PPCPs (carbamazepine

and propranolol) in contaminated soil with the addition of mallee eucalyptus oil and wheat chaff-derived biochar. Table 7 lists the important findings of studies on biochar remediation on PPCPs in different vegetables; more studies are encouraged involving the interaction of biochar and PPCPs in different vegetables in the near future.

8.3. The Impact of Biochar on the Growth of Vegetables in soil Contaminated with TE and PPCP Contaminants

The growth performance of vegetables has shown a significant increase in yield and biomass after the addition of biochar [174]. For instance, the effect of biochar on the growth and yield of tomatoes grown in TE-contaminated soil was investigated in a study [175]. The maize stalk-derived biochar showed a decrease in the uptake of Cu, Cd, and Pb in tomatoes. Moreover, it also resulted in a significant rise in the quality, yields, and photosynthetic pigments of the tomatoes. The authors concluded that the amendment with biochar showed a positive impact on the growth and productivity of the tomato along with a reduction in the TEs concentration in a two-year field study [175]. The impact of biochar was investigated in another study in lettuces which were grown in Cr-contaminated soil [176]. The results showed an increase in the nutrient uptake and a decrease in the Cr of the lettuces. Li et al. [177] also found a considerable rise in the biomass and yield, by 30–65% more than controls, for pakchoi cabbages grown on multi-element-contaminated soil with As, Cd, and Cu after amendment with different doses of biochar. Similar results were found in different vegetable species such as spinach [178], eggplant [179], lettuce [180], tomato [181], and carrot [182]. In addition, the effect of biochar application on vegetables grown in PPCP-contaminated soils have also shown similar results. For instance, in a study by Li et al. [20], biochar amendment showed a positive effect on the reduction and yield of radishes contaminated with PPCPs. Biochar application has shown to increase root length, and plant biomass and yield with the correct or appropriate quantity of biochar dose; however, it may vary between the contaminant type, variety of vegetable, and the biochar dose applied. For example, the effect of walnut shell-derived biochar was studied in lettuces and carrots grown in PPCP-contaminated soil [167]. An increase in the root and shoot biomass of the lettuce was observed after application of the walnut shell-derived biochar. In contrast no difference in root and shoot biomass was observed for carrots grown in ciprofloxacin, triclosan, and triclocarbon-contaminated soil [167].

9. Conclusions and Future Outlook

Contamination by TEs and ECs such as PPCPs is a matter of concern for soil–vegetable systems. The exponential growth of the human population, industrialization, and urbanization have introduced large varieties of chemicals into the soil and surrounding environment. Therefore, the peculiar rise of such contamination is considered to be a risk. This article reviewed the potential contaminants within the traditional and emerging contaminants in soil–vegetable systems, and their sources, accumulation, uptake factors, health risks, and mitigation strategies using biochar. TEs (As, Pb, Cu, Cr, Cd) and PPCPs are the major groups of contaminants in soil which affect the soil–vegetable interface adversely. Adequate measures should be taken for mitigating TE and PPCP contamination in soil in order to reduce the potential risk in humans and the environment. Some recommended strategies are as follows:

1. Vegetable crops should be grown in a contaminant-free environment with limited/regulated fertilizers and pesticides;
2. Better initiatives and awareness programs should be initiated frequently by the local community and government for addressing soil contaminants and their associated health risk among the common man and the farmers;
3. VTE and PPCP discharges that are released from individual, industrial, and agricultural activities should be regulated and monitored accurately;
4. More field trials involving the application of biochar for soil remediation should be carried out;

5. VMonitoring the production of contaminant-free biochar is also necessary before applying biochar for soil remediation.

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Abbreviations

TEs	trace elements
ECs	emerging contaminants
PPCPs	pharmaceuticals and personal care products
Cd	Cadmium
Cu	Copper
Pb	Lead
As	Arsenic
Si	Silicon
Hg	Mercury
Co	Cobalt
Cr	Chromium
CEC	cation exchange capacity
Ca	Calcium
Mg	Magnesium
N	Nitrogen
P	Phosphorous
Zn	Zinc
Mo	Molybdenum
Fe	Iron
Ni	Nickel
F	Fluorine
ROS	reactive oxygen species
Eh	redox potential
YS	yellow stripe
HMAAs	heavy metal transporting ATPase
FPN	Ferroportin
COPT	copper transporter
CAX	cation exchanger
CO ₂	carbon di oxide
BCF	bio concentration factor
RCF	root concentration factor
TF	translocation factor

HQ	hazard quotient
THQ	target hazard quotient
HRI	health risk index
DIM	daily intake of metal
WBCs	white blood cells
RQ	risk quotient
HHI	health hazard index
EDI	estimated daily intake
ADI	accepted daily intake
XRD	X-ray diffraction
SEM	scanning electron microscopy
FTIR	Fourier transform infrared

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