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# Characteristics of HfO<sub>2</sub> thin films deposited by plasma-enhanced atomic layer deposition using O<sub>2</sub> plasma and N<sub>2</sub>O plasma

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The characteristics of HfO<sub>2</sub> dielectrics deposited by the plasma-enhanced atomic layer deposition (PEALD) method using O<sub>2</sub> and N<sub>2</sub>O plasmas were investigated. The deposited HfO<sub>2</sub> films had a randomly oriented polycrystalline phase while the interfacial layers of the films were amorphous. During the PEALD process with N<sub>2</sub>O plasma, nitrogen was mainly incorporated into the interfacial region between the HfO<sub>2</sub> film and the Si substrate. The nitrogen content of 2–3 at. % in the interface was analyzed by Auger electron spectroscopy. The incorporated nitrogen at the interface effectively suppressed residual oxygen diffusion during subsequent annealing at 800 °C in a N<sub>2</sub> atmosphere. A thicker interfacial layer was observed in the as-deposited and annealed HfO<sub>2</sub> film with O<sub>2</sub> plasma than with N<sub>2</sub>O plasma. For HfO<sub>2</sub> films prepared with the N<sub>2</sub>O plasma, where equivalent oxide thickness (EOT) increased from 1.43 to 1.56 nm after annealing, the leakage current densities, measured at a gate bias voltage of  $|V_G - V_{FB}| = 2$ , increased from  $3.5 \times 10^{-8}$  to  $4.8 \times 10^{-8}$  A/cm<sup>2</sup>. For HfO<sub>2</sub> films prepared with the O<sub>2</sub> plasma, where EOT increased from 1.60 to 2.01 nm after annealing, the leakage current densities decreased from  $1.1 \times 10^{-6}$  to  $1.3 \times 10^{-7}$  A/cm<sup>2</sup>. The film with O<sub>2</sub> plasma had a higher amount of negative fixed oxide charges than the film with N<sub>2</sub>O plasma. N<sub>2</sub>O plasma improved the leakage current properties by allowing nitrogen incorporation at the interfacial region and less crystallization of HfO<sub>2</sub> film.

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## I. INTRODUCTION

Conventional SiO<sub>2</sub> gate dielectrics suffer from leakage and reliability deficiencies as the SiO<sub>2</sub> thickness in the metal-oxide-semiconductor field effect transistors (MOSFETs) decreases below 2.0 nm.<sup>1</sup> One solution to effectively reduce the equivalent oxide thickness (EOT) without increasing the leakage current is to use high dielectric constant (high-*k*) materials.<sup>2</sup> This application of high-*k* gate dielectrics to replace SiO<sub>2</sub> is now actively investigated. Metal-organic chemical vapor deposition (MOCVD) and atomic layer deposition (ALD) have been widely studied in an effort to grow high-*k* gate dielectrics.<sup>3,4</sup> These deposition methods exhibit the good uniformity and damage-free film characteristics compared to physical vapor deposition (PVD) methods. However, contaminations such as hydrogen, halogens, and organic species can deteriorate the electrical properties of films grown at low temperatures by MOCVD and ALD methods.<sup>5,6</sup> Therefore, the plasma application in the ALD process is one way to solve these problems because plasma effects increase reactivity, reduce impurities, and densify the film.<sup>7</sup> Among the high dielectric constant materials, HfO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>, and ZrO<sub>2</sub> have been widely studied as next-generation semiconductor devices that solve the problems of excessive high leakage current.<sup>8,9</sup> HfO<sub>2</sub> is considered to be one of the most promising materials among these high di-

electric materials. HfO<sub>2</sub> exhibits desirable properties such as high dielectric constant ( $\kappa = 25\text{--}30$ ), high density ( $\sim 9.65$  g/cm<sup>3</sup>), large band gap (5.68 eV), and thermal stability in contact with silicon.<sup>10</sup> Unfortunately, HfO<sub>2</sub> crystallizes at temperatures less than 500 °C.<sup>11</sup> Grain boundaries found in crystallized HfO<sub>2</sub> can be used as diffusion paths for oxygen or dopants through the oxide layer, causing a threshold voltage instability, and the generation of defects.<sup>12</sup> The formation of an uncontrolled interfacial layer at the HfO<sub>2</sub>/Si interface also limits further scaling down of EOT of HfO<sub>2</sub> gate dielectric. HfO<sub>2</sub> films have also shown a poor thermal stability after subsequent thermal processes, which results in an increase of leakage current. However, several research groups have recently reported that the incorporation of nitrogen into high-*k* materials can improve the material's thermal stability.<sup>13,14</sup> Incorporation of nitrogen atoms into the HfO<sub>2</sub> films during deposition would prevent crystallization of the HfO<sub>2</sub> film and interfacial reactions between the HfO<sub>2</sub> and Si after annealing.

In this study, we utilized the plasma-enhanced atomic layer deposition (PEALD) system and focused on the effect of the incorporated nitrogen in HfO<sub>2</sub> films during PEALD process when using N<sub>2</sub>O plasma. The characteristics of the HfO<sub>2</sub> films deposited by PEALD using N<sub>2</sub>O plasma were then compared to that of the films deposited using O<sub>2</sub> plasma.

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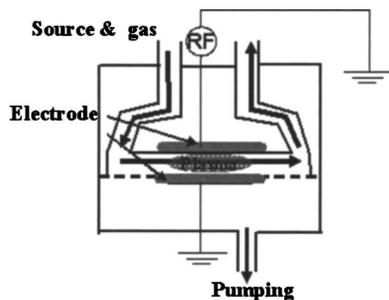


FIG. 1. Schematic of the PEALD system.

## II. EXPERIMENT

The HfO<sub>2</sub> films were deposited on *p*-type Si (100) wafers at 340 °C by the PEALD method using O<sub>2</sub> or N<sub>2</sub>O plasma with tetrakis-diethyl-amino-hafnium [Hf(N(C<sub>2</sub>H<sub>5</sub>)<sub>2</sub>)<sub>4</sub>], as the Hf precursor. Si wafers were cleaned with a piranha solution (H<sub>2</sub>O<sub>2</sub>:H<sub>2</sub>SO<sub>4</sub>=1:4) for 10 min to remove organic contaminants, washed in de-ionized (DI) water, and then dipped for 2 min in a dilute HF solution (HF:DI water=1:100) to remove native oxides. The clean wafers were then rinsed in DI water again, dried in nitrogen, and loaded into the ALD chamber. Figure 1 shows the schematic drawing of PEALD system. The Hf precursor, Hf(N(C<sub>2</sub>H<sub>5</sub>)<sub>2</sub>)<sub>4</sub>, was vaporized at 90 °C and introduced into the reaction chamber by using a bubbler with Ar as the carrier gas. The O<sub>2</sub> plasma and N<sub>2</sub>O plasma were used as the oxygen reactants. The basic one cycle consisted of supplying the Hf precursor and then exposing the plasma. Ar purge gas was introduced to promote the complete separation of the Hf precursor and plasma. All gas flows were automatically controlled by using a solenoid valve. For the HfO<sub>2</sub> film, the sequential processing time for the Hf(N(C<sub>2</sub>H<sub>5</sub>)<sub>2</sub>)<sub>4</sub>, Ar purge, plasma, and Ar purge were 5, 7, 5, and 7 s, respectively. The plasma source used was a radio-frequency (rf) source of 13.56 MHz and a fixed power of 100 W. The process pressure was maintained at 0.5 torr by automatic control with a throttle valve. The metal-oxide-semiconductor (MOS) capacitors were fabricated with an ~5 nm thick HfO<sub>2</sub> film and a Pt electrode with a thickness

of ~100 nm. For postdeposition annealing (PDA), the films were rapidly annealed at 800 and 1000 °C for 1 min in a N<sub>2</sub> atmosphere. A postmetallization annealing (PMA) was also carried out in a 97% N<sub>2</sub>/3% H<sub>2</sub> mixed atmosphere at 450 °C for 30 min. The physical thickness of the HfO<sub>2</sub> films and interfacial layers were analyzed by cross-sectional transmission electron microscopy (XTEM). The chemical states and bonding structure of the HfO<sub>2</sub> films were investigated using x-ray photoelectron spectroscopy (XPS). Auger electron spectroscopy (AES) was utilized to analyze variations in chemical composition, the content of the impurities, and the amount of incorporated nitrogen. The electrical properties and reliability characteristics including EOT, hysteresis, capacitance, and leakage current were measured using a Keithley 590 *C-V* analyzer and an HP 4155A semiconductor parameter analyzer.

## III. RESULTS AND DISCUSSION

Figure 2 shows the AES depth profiles of as-grown HfO<sub>2</sub> films using (a) O<sub>2</sub> plasma and (b) N<sub>2</sub>O plasma at a deposition temperature of 340 °C. The AES depth profile of both samples shows the stoichiometric composition of the HfO<sub>2</sub> films. The carbon contents in the HfO<sub>2</sub> films deposited with O<sub>2</sub> plasma and N<sub>2</sub>O plasma were approximately 1–2 at. %. As shown in Fig. 2(b), the nitrogen content at the interfacial region of the HfO<sub>2</sub> films deposited with the N<sub>2</sub>O plasma was approximately 2–3 at. %, while nitrogen atoms were not detected in the HfO<sub>2</sub> film deposited with the O<sub>2</sub> plasma, as seen in Fig. 2(a). The nitrogen atoms were incorporated not in HfO<sub>2</sub> film but at the interfacial region of the film during the PEALD process using the N<sub>2</sub>O plasma. Lucovsky reported that nitrogen is incorporated at the Si–SiO<sub>2</sub> interface and not in the bulk of the thin SiO<sub>2</sub> in the remote-plasma-assisted oxidation (RPAO) process using N<sub>2</sub>O.<sup>15</sup> The AES result of Fig. 2(b) shows that the nitrogen is incorporated at the Si–HfO<sub>2</sub> interface during HfO<sub>2</sub> deposition using N<sub>2</sub>O plasma.

Figure 3 shows XPS spectra of the Hf 4*f* for the as-deposited and annealed ~5 nm thick HfO<sub>2</sub> films with O<sub>2</sub>

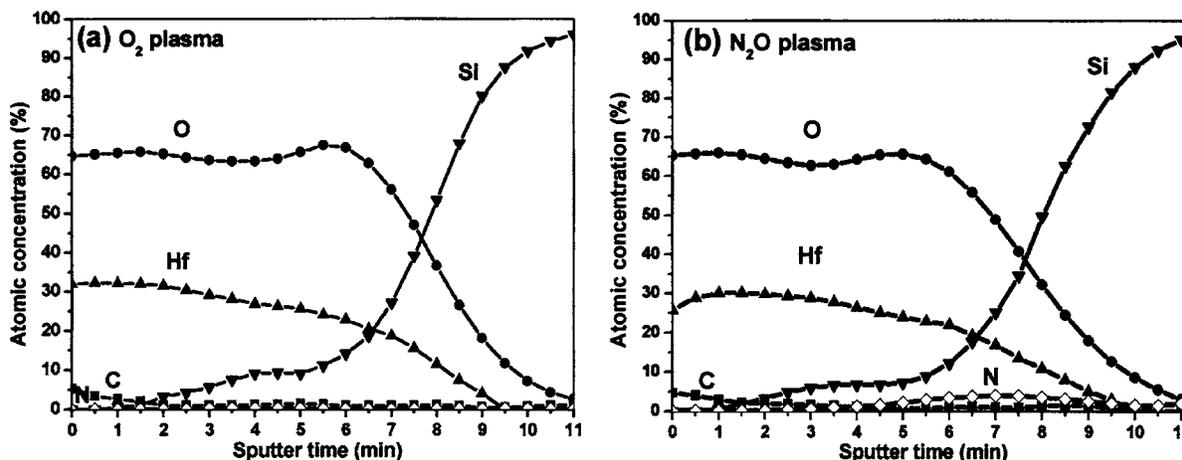


FIG. 2. AES depth profiles of as-deposited HfO<sub>2</sub> films using (a) O<sub>2</sub> plasma and (b) N<sub>2</sub>O plasma at a deposition temperature of 340 °C.

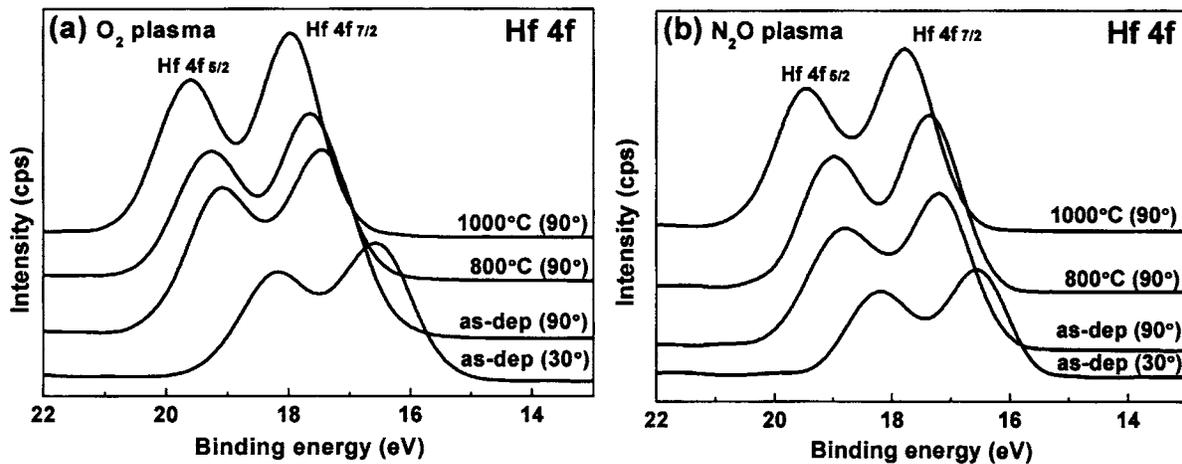


FIG. 3. XPS spectra of the Hf 4f for the  $\sim 5$  nm thick HfO<sub>2</sub> films after a PDA at temperatures of 800 and 1000 °C for 1 min in a N<sub>2</sub> atmosphere: (a) HfO<sub>2</sub> films deposited with O<sub>2</sub> plasma and (b) HfO<sub>2</sub> films deposited with N<sub>2</sub>O plasma. The as-deposited films were measured at take-off angles of 30° and 90°.

plasma and N<sub>2</sub>O plasma. The PDA was performed at temperatures of 800 and 1000 °C for 1 min in a N<sub>2</sub> atmosphere. The as-deposited films were measured at take-off angles of 30° and 90°. At take-off angle of 30°, the signal from the HfO<sub>2</sub> film becomes greatly enhanced relative to that from the interfacial region between HfO<sub>2</sub> and Si. At take-off angle of 90°, the signal at the interface of HfO<sub>2</sub> and Si becomes enhanced. We calibrated all the peaks with the C 1s peak position at the 285 eV. Wilk *et al.* reported that the Hf 4f<sub>7/2</sub> peak of Hf silicate is  $\sim 1.0$  eV higher than that of the HfO<sub>2</sub>, which is located at  $\sim 16.5$ – $17$  eV.<sup>16</sup> As shown in the Hf 4f<sub>7/2</sub> peaks of Fig. 3, as-deposited HfO<sub>2</sub> films with O<sub>2</sub> or N<sub>2</sub>O plasma exhibit a binding energy of  $\sim 16.5$  eV at take-off angle of 30°. At take-off angle of 90°, Hf 4f<sub>7/2</sub> peaks of as-deposited HfO<sub>2</sub> films with O<sub>2</sub> and N<sub>2</sub>O plasmas are located at  $\sim 17.4$  and  $\sim 17.0$  eV, respectively. Therefore, the films are considered to consist of HfO<sub>2</sub> and Hf silicate, where the Hf silicate is in the interfacial layer. The nitrogen incorporated in the interface by the N<sub>2</sub>O plasma in Fig. 2(b) may also contribute to the peak shift toward a lower binding

energy than Hf 4f<sub>7/2</sub> peaks position of HfO<sub>2</sub> film deposited with O<sub>2</sub> plasma in Fig. 3. When the annealing temperatures of HfO<sub>2</sub> samples were increased, the Hf 4f<sub>7/2</sub> peak at take-off angle of 90° shifted from  $\sim 17.4$  to  $\sim 18.0$  eV for O<sub>2</sub> plasma and from  $\sim 17.1$  to  $\sim 17.8$  eV for N<sub>2</sub>O plasma. The Hf 4f<sub>7/2</sub> peak for O<sub>2</sub> plasma shifted to a higher binding energy, as compared with N<sub>2</sub>O plasma. This indicates that the nitrogen incorporated into the interface of HfO<sub>2</sub> film by the N<sub>2</sub>O plasma effectively suppressed the growth of the Hf silicate layer.

Figures 4(a) and 4(b) show the XPS spectra of the Si 2p of the  $\sim 5$  nm thick HfO<sub>2</sub> films deposited with O<sub>2</sub> plasma and N<sub>2</sub>O plasma, respectively, after the PDA at 800 and 1000 °C for 1 min in a N<sub>2</sub> atmosphere. The HfO<sub>2</sub> films were measured at take-off angle of 90° to observe the growth of interfacial layer with annealing temperature. The Si 2p peak at  $\sim 99$  eV originated from a Si substrate. The binding energies observed between  $\sim 102$  and  $\sim 103.3$  eV corresponded to the interfacial layer. As the annealing temperatures increased, the intensities of the XPS peaks corresponding to

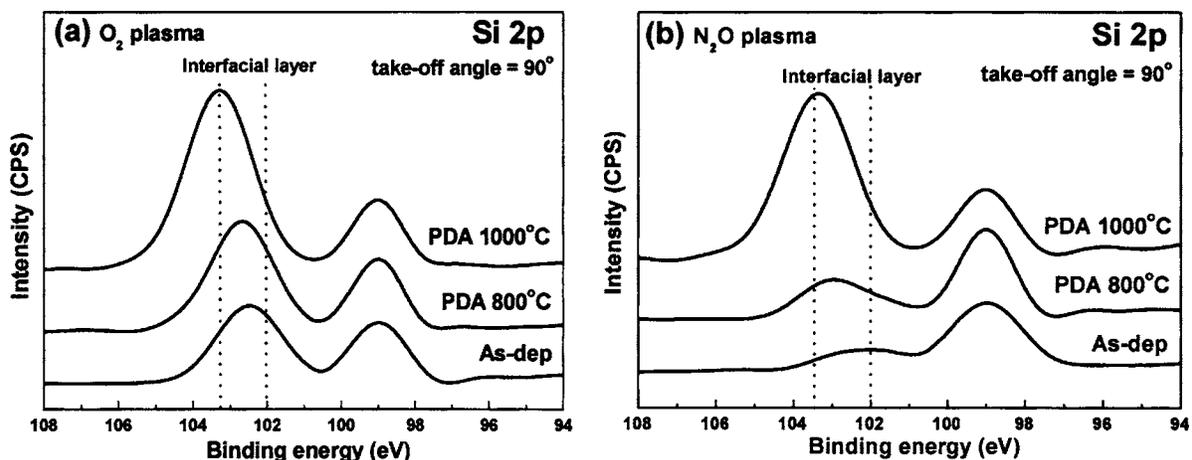


FIG. 4. XPS spectra of Si 2p for the  $\sim 5$  nm thick HfO<sub>2</sub> films after a PDA at the temperatures of 800 and 1000 °C for 1 min in a N<sub>2</sub> atmosphere: (a) HfO<sub>2</sub> films deposited with O<sub>2</sub> plasma and (b) HfO<sub>2</sub> films deposited with N<sub>2</sub>O plasma. The films were measured at take-off angle of 90°.

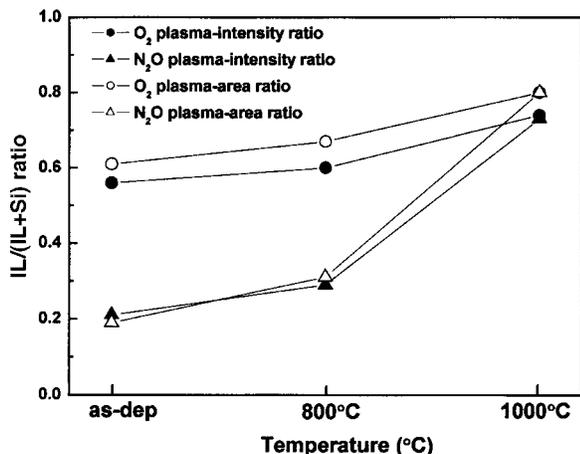


FIG. 5. Intensity and area ratios of Si 2p peaks measured at take-off angle of 90° with varying PDA temperatures. The intensity and area ratios,  $IL/(IL+Si)$ , were calculated by fitting with a Gaussian function.

the interfacial layer increased, which corresponded to a growth of the interfacial layer. However, the interfacial layers of HfO<sub>2</sub> films with the N<sub>2</sub>O plasma were thinner than those with the O<sub>2</sub> plasma up to a temperature of 800 °C. This indicates that the incorporated N atoms into the interfacial region by the N<sub>2</sub>O plasma suppressed the growth of the interfacial layer.

Figure 5 shows the intensity and area ratio of Si 2p peaks measured at take-off angles of 90° varying with PDA temperatures. The intensity and area ratio of  $IL/(IL+Si)$  were calculated by fitting with a Gaussian function. The IL and Si represent the interfacial layer and silicon substrate, respectively. As the annealing temperature increased, the intensity and area ratio of  $IL/(IL+Si)$  increased. The intensity and area ratio of  $IL/(IL+Si)$  for N<sub>2</sub>O plasma was lower than that for O<sub>2</sub> plasma up to a temperature of 800 °C, while the two were almost identical at 1000 °C after a sharp increase from 800 to 1000 °C. This shows that the interfacial layer formed by N<sub>2</sub>O plasma is thinner than that formed by O<sub>2</sub> plasma up to temperatures of 800 °C. Therefore, the nitrogen incorporated into the interfacial layer acts as a reaction-diffusion barrier during the PEALD process.

Four images of Fig. 6 show cross-sectional TEM of ap-

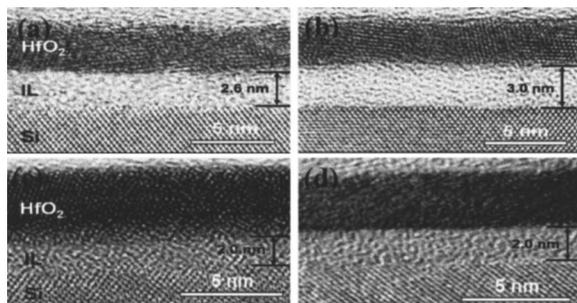


FIG. 6. Cross-sectional TEM images of approximately 5.5 nm HfO<sub>2</sub> deposited films. (a) is an as-deposited sample with O<sub>2</sub> plasma and (b) is an annealed sample at 800 °C for 1 min. (c) is an as-deposited sample with N<sub>2</sub>O plasma and (d) is an annealed sample at 800 °C for 1 min.

proximately 5.5 nm HfO<sub>2</sub> deposited films. Figure 6(a) is an as-deposited sample with O<sub>2</sub> plasma and Fig. 6(b) is an annealed sample at 800 °C for 1 min. Figure 6(c) is an as-deposited sample with N<sub>2</sub>O plasma and Fig. 6(d) is an annealed sample at 800 °C for 1 min. As shown in Fig. 6, the as-deposited and the annealed HfO<sub>2</sub> films exhibited a randomly oriented crystalline structure and their interfacial layers had an amorphous structure. We think that the metastable species in the plasma released their energy through collisions, resulting in the observed crystallization.<sup>17</sup> Therefore, the crystallization of the HfO<sub>2</sub> film grown by the PEALD method was considered to be caused by the metastable species in plasma which enhanced the reaction. The ions in the plasma physically attacked the film, and radicals chemically reacted with ligands of the Hf(N(C<sub>2</sub>H<sub>5</sub>)<sub>2</sub>)<sub>4</sub> precursor during the PEALD process. The degree of crystallization of the HfO<sub>2</sub> film with O<sub>2</sub> plasma in Figs. 6(a) and 6(b) is larger compared to that with N<sub>2</sub>O plasma in Figs. 6(c) and 6(d). The interfacial layers of HfO<sub>2</sub> film with O<sub>2</sub> plasma were thicker than those with the N<sub>2</sub>O plasma, which agrees well with the XPS results in Fig. 5. The interfacial layer thickness of the HfO<sub>2</sub> film deposited using O<sub>2</sub> plasma increased from 2.6 to 3.0 nm, while those using N<sub>2</sub>O plasma remained at a constant thickness of ~2.0 nm after PDA at 800 °C for 1 min. The thin interfacial layers in Figs. 6(c) and 6(d) are due to the incorporated nitrogen atoms in the interfacial region by the N<sub>2</sub>O plasma. Therefore, nitrogen atoms incorporated in the interfacial region from the N<sub>2</sub>O plasma, effectively suppressing the diffusion of residual oxygen and/or silicon into the film during deposition or annealing process and contributing to the suppression of the interfacial layer growth.

Figure 7 shows the capacitance-voltage (*C-V*) curves of the HfO<sub>2</sub> films grown with (a) O<sub>2</sub> plasma and (b) N<sub>2</sub>O plasma after an annealing treatment at 800 °C for 1 min. The *C-V* was measured at a frequency of 100 kHz. The gate voltage was swept from accumulation to inversion and back. The *C-V* curves showed maximum accumulation at high negative bias voltages. The corresponding EOT values of the as-deposited HfO<sub>2</sub> films with O<sub>2</sub> plasma and N<sub>2</sub>O plasma were approximately 1.60 and 1.43 nm, respectively, and those of the 800 °C annealed films were 2.0 and 1.56 nm, respectively. The increase of EOT is related to the interfacial layer growth and the increase in the total film thickness after annealing. The capacitance of the as-deposited HfO<sub>2</sub> film in the accumulation region decreased after annealing. This resulted from the decrease in the series capacitance due to the interfacial layer growth. Therefore, from a variation of capacitances in accumulation region of the *C-V* curves in Fig. 7, the nitrogen-incorporated HfO<sub>2</sub> films with the N<sub>2</sub>O plasma suppressed the growth of the interlayer more effectively than the film deposited by O<sub>2</sub> plasma. This was also confirmed by the thickness of the interfacial layer that was obtained from high-resolution TEM (HRTEM) in Fig. 6. The flatband voltage of the *C-V* curves was shifted toward positive bias during sweeping. Since the electrons injected under a positive bias are not ejected rapidly, they induce a negative charge at

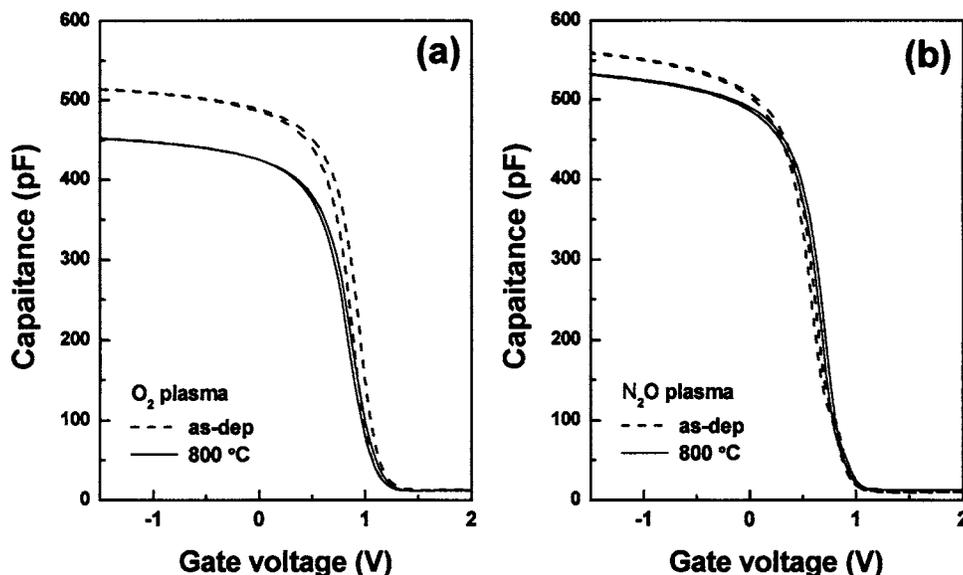


FIG. 7. Capacitance-voltage ( $C$ - $V$ ) curves of HfO<sub>2</sub> films grown using O<sub>2</sub> plasma and N<sub>2</sub>O plasma after an annealing treatment at 800 °C for 1 min.

the interface of the HfO<sub>2</sub> films. The HfO<sub>2</sub> films showed a reversible hysteresis indicating that there is no significant effect from the deep traps in the bulk. For the HfO<sub>2</sub> film with O<sub>2</sub> plasma, the flatband voltage ( $V_{FB}$ ) and effective fixed oxide charge density ( $Q_{f,eff}$ ) decreased from 1.23 to 1.18 and from  $-6.01 \times 10^{12}$  to  $-4.84 \times 10^{12}$  q/cm<sup>2</sup>, respectively, as a result of the annealing process. For those with N<sub>2</sub>O plasma,  $V_{FB}$  and  $Q_{f,eff}$  decreased from 0.91 to 0.89 and from  $-2.95 \times 10^{12}$  to  $-2.57 \times 10^{12}$  q/cm<sup>2</sup>, respectively. Therefore, the film with O<sub>2</sub> plasma had a higher amount of negative fixed oxide charges than those with N<sub>2</sub>O plasma.

Figure 8 shows the current-voltage ( $J$ - $V$ ) curves of the HfO<sub>2</sub> films grown by O<sub>2</sub> plasma and N<sub>2</sub>O plasma after an annealing treatment at 800 °C for 1 min. The leakage current densities of the as-deposited and annealed HfO<sub>2</sub> films measured at a gate bias voltage of  $|V_G - V_{FB}| = 2$  were  $1.1 \times 10^{-6}$  and  $1.3 \times 10^{-7}$  A/cm<sup>2</sup> in the case of the O<sub>2</sub> plasma and  $3.5 \times 10^{-8}$  and  $4.8 \times 10^{-8}$  A/cm<sup>2</sup> in the case of the N<sub>2</sub>O plasma, respectively. The reduction of leakage current density of the HfO<sub>2</sub> films deposited using O<sub>2</sub> plasma after an-

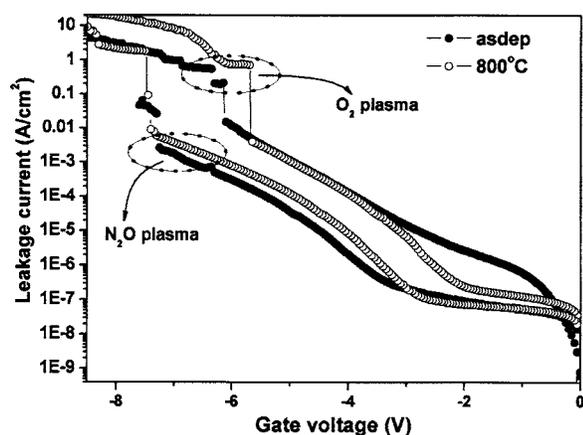


FIG. 8. Current-voltage ( $J$ - $V$ ) curves of HfO<sub>2</sub> films grown using O<sub>2</sub> plasma and N<sub>2</sub>O plasma after an annealing treatment at 800 °C for 1 min.

nealing is believed to be due to changes in the growth of the amorphous interfacial layer and the film densification caused by postdeposition annealing. The nitrogen incorporation at the interfacial region and less crystallization due to the N<sub>2</sub>O plasma resulted in an improvement of the leakage current properties. Table I summarizes the electrical characteristics of the HfO<sub>2</sub> films deposited by using PEALD.

#### IV. CONCLUSIONS

HfO<sub>2</sub> films were successfully deposited on Si substrates using Hf(N(C<sub>2</sub>H<sub>5</sub>)<sub>2</sub>)<sub>4</sub> as a precursor with O<sub>2</sub> plasma and N<sub>2</sub>O plasma. The nitrogen atoms were incorporated at the interfacial region of the HfO<sub>2</sub> film during the PEALD process using N<sub>2</sub>O plasma. The interfacial layer of HfO<sub>2</sub> films with N<sub>2</sub>O plasma was thinner than films with O<sub>2</sub> plasma up to a temperature of 800 °C. HfO<sub>2</sub>/IL/Si structure deposited by N<sub>2</sub>O plasma was shown to be more resistant to oxygen diffusion. The nitrogen atoms incorporated into the interface of the HfO<sub>2</sub> film by the N<sub>2</sub>O plasma effectively suppressed the growth of the interfacial Hf silicate layer. The HfO<sub>2</sub> films deposited with the N<sub>2</sub>O plasma also showed improved film quality with a relatively low EOT, low leakage current density, and low effective fixed oxide charge density compared

TABLE I. Summary of the electrical characteristics of HfO<sub>2</sub> films deposited by PEALD using O<sub>2</sub> plasma and N<sub>2</sub>O plasma. LKG represents leakage current density measured at  $|V_G - V_{FB}| = 2$ .  $V_{FB}$  and  $Q_{f,eff}$  represents flatband voltage and effective fixed oxide charge density, respectively.

Oxygen reactant	Annealing temperature (°C)	EOT (nm)	LKG (A cm <sup>-2</sup> )	$V_{FB}$ (V)	$Q_{f,eff}$ (q cm <sup>-2</sup> )
O <sub>2</sub> plasma	As deposited	1.60	$1.1 \times 10^{-6}$	1.23	$-6.01 \times 10^{12}$
	800	2.01	$1.3 \times 10^{-7}$	1.18	$-4.84 \times 10^{12}$
N <sub>2</sub> O plasma	As deposited	1.43	$3.5 \times 10^{-8}$	0.91	$-2.95 \times 10^{12}$
	800	1.56	$4.8 \times 10^{-8}$	0.89	$-2.57 \times 10^{12}$

to films deposited with the O<sub>2</sub> plasma. Therefore, the N<sub>2</sub>O plasma process can improve the physical and electrical properties of HfO<sub>2</sub> film.

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