



Article

Compact SnO₂/Mesoporous TiO₂ Bilayer Electron Transport Layer for Perovskite Solar Cells Fabricated at Low Process Temperature

Junyeong Lee ¹, Jongbok Kim ², Chang-Su Kim ³ and Sungjin Jo ^{1,*}

¹ School of Energy Engineering, Kyungpook National University, Daegu 41566, Korea; junyeong112@knu.ac.kr

² Department of Materials Science and Engineering, Kumoh National Institute of Technology, Gumi 39177, Korea; jbkim@kumoh.ac.kr

³ Department of Advanced Functional Thin Films, Surface Technology Division, Korea Institute of Materials Science, Changwon 51508, Korea; cskim1025@kims.re.kr

* Correspondence: sungjin@knu.ac.kr

Abstract: Charge transport layers have been found to be crucial for high-performance perovskite solar cells (PSCs). SnO₂ has been extensively investigated as an alternative material for the traditional TiO₂ electron transport layer (ETL). The challenges facing the successful application of SnO₂ ETLs are degradation during the high-temperature process and voltage loss due to the lower conduction band. To achieve highly efficient PSCs using a SnO₂ ETL, low-temperature-processed mesoporous TiO₂ (LT m-TiO₂) was combined with compact SnO₂ to construct a bilayer ETL. The use of LT m-TiO₂ can prevent the degradation of SnO₂ as well as enlarge the interfacial contacts between the light-absorbing layer and the ETL. SnO₂/TiO₂ bilayer-based PSCs showed much higher power conversion efficiency than single SnO₂ ETL-based PSCs.

Keywords: compact SnO₂; mesoporous TiO₂; oxygen plasma; perovskite solar cell low process temperature



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1. Introduction

Perovskite solar cells (PSCs) have received attention because their power conversion efficiency (PCE) has rapidly increased by over 25% [1,2]. Many researchers have tried to enhance the performance of PSCs and translate them from the laboratory to commercial products [3–6].

For the state-of-the-art device configuration, PSCs usually consist of a transparent electrode, an electron transport layer (ETL), a light-absorbing layer, a hole transport layer (HTL), and a metal electrode [7]. In the pursuit of high-performance PSCs, the ETL has become the subject of high interest and one of the most challenging scientific issues [8]. As a conventional ETL material, TiO₂ has been widely adopted. However, many attempts have been made to substitute TiO₂ with alternative materials that have better optoelectronic properties [9]. SnO₂ is the most investigated ETL after TiO₂ due to its high electron mobility, high conductivity, wide optical bandgap, and excellent chemical stability [10]. Although SnO₂ ETL-based PSCs have made rapid progress recently, their performance is still lower than that of PSCs using mesoporous TiO₂ (m-TiO₂) as an ETL [11]. In addition, the low conduction band of SnO₂ reduces the built-in potential of the Schottky barrier between the perovskite and SnO₂, resulting in the voltage loss of the PSCs [12].

To achieve highly efficient PSCs using a SnO₂ ETL, SnO₂ ETL combination and surface modification techniques that can improve electron injection and suppress electron recombination have been developed [13]. Various inorganic metal oxides, such as ZnO [14], MgO [15], and TiO₂ [16,17], as well as organics, including carbon-based materials [18], self-assembled monolayers (SAM) [19], and polymers [20], have been adopted in SnO₂

ETL-based PSCs to combine with or modify SnO₂. Among them, the conventional m-TiO₂ layer is the preferable candidate to be combined with the compact SnO₂ (c-SnO₂) layer because the mesoporous scaffold can facilitate sufficient pore filling of the light-absorbing layer and improve electron extraction and transport over a single c-SnO₂ layer [17,21]. Moreover, TiO₂ is a better choice in view of its established cascaded energy-level alignment between the electrode and light-absorbing layer, which results in a significantly improved performance. However, m-TiO₂ generally requires a high-temperature sintering process of up to 450 °C to remove organic additives that cause deterioration in the photovoltaic performance [22]. This high-temperature process restricts the application of m-TiO₂ on the c-SnO₂ layer because the high-temperature process induces not only a large amount of charge traps and a recombination center in the SnO₂ layer but also poor interfacial contact, leading to interface recombination and shunting paths. Therefore, one important challenge is determining how to construct m-TiO₂ on the c-SnO₂ layer to take advantage of SnO₂. We recently achieved low-temperature processed PSCs by employing m-TiO₂ as ETL [23]. To remove the organic additives in the low-temperature-processed TiO₂ (LT-TiO₂), we adopted the oxygen plasma process. The simple and effective method of oxygen plasma treatment enhances charge extraction and transport, thereby improving photovoltaic performance. Therefore, our newly developed oxygen plasma treatment for LT m-TiO₂ is a promising strategy for combining the m-TiO₂ layer with the c-SnO₂ layer to produce an efficient bilayer ETL.

In this work, we demonstrated that the LT m-TiO₂ can be adopted to construct a compact/mesoporous structured bilayer ETL to prevent the degradation of SnO₂ by the high-temperature process. When the conventional m-TiO₂ layer was deposited on the SnO₂ layer and then the bilayer ETL underwent the high-temperature sintering process (BLH), the photovoltaic performance of this bilayer ETL-based PSC (BLH-PSC) deteriorated more than that of a single c-SnO₂ ETL-based PSC (SL-PSC). On the contrary, when the oxygen plasma treatment was applied to the LT m-TiO₂ deposited on the SnO₂ (BLP), the PSC with this bilayer ETL (BLP-PSC) exhibited an excellent PCE of 15.36%, which is higher than that of the SL-PSC (13.68%). Moreover, detailed characterizations demonstrated that the SnO₂/TiO₂ bilayer ETL is beneficial for carrier extraction and transport.

2. Materials and Methods

2.1. Fabrication of the SnO₂ Layer

A fluorine-doped tin oxide (FTO) electrode was patterned using zinc powder and diluted HCl solution. Then, the patterned FTO substrate was cleaned with deionized water (DI), acetone, and ethanol in an ultrasonic bath. After ultraviolet–ozone (UVO) treatment for 15 min, 0.05 M SnCl₂·2H₂O solution diluted in ethanol was spin-coated on the patterned FTO substrate and sintered at 200 °C for 1 h.

2.2. Fabrication of the Bilayer ETL

The m-TiO₂ solution was prepared by dissolving TiO₂ nanoparticle paste (Dyesol, Queanbeyan, Australia) in ethanol at a ratio of 1:10 (wt %). After 15 min of UVO treatment, the m-TiO₂ solution was spin-coated on the SnO₂ layer and sintered at 150 °C for 4 h for BLL-PSC and BLP-PSC. To remove TiO₂ nanoparticle aggregates, the bilayer ETL substrate was dipped in ethanol and stirred for 15 s, and then the substrate was annealed at 150 °C for 30 min. In contrast, the m-TiO₂ layer was sintered at 450 °C for 1 h and was not rinsed with ethanol for BLH-PSC. The substrate was dipped in 20 mM TiCl₄ solution at 90 °C for 15 min and sintered at 150 °C for 30 min.

2.3. Fabrication of the PSC

The MAPbI₃ solution was prepared by mixing methylammonium iodide (MAI), PbI₂, dimethyl sulfoxide, and *N,N*-dimethylformamide. In the case of BLP-PSCs, the m-TiO₂ layer was treated by oxygen plasma at a radio frequency (RF) power of 20 W for 10 min. Then, the MAPbI₃ layer was spin-coated and annealed at 65 °C for 1 min and at 100 °C for

10 min. The mixed Spiro-OMeTAD solution, which contained Spiro-OMeTAD (Jilin OLED, Changchun, Jilin Sheong, China), lithium salt, 4-tert-butylpyridine, and chlorobenzene, was spin-coated on the MAPbI₃ layer. A silver electrode was deposited via a thermal evaporator. All chemicals were purchased from Sigma-Aldrich (St. Louis, MO, USA).

2.4. Fabrication of the Flexible PSC

The indium-doped tin oxide (ITO)/polyethylene naphthalate (PEN) substrate (Pec-cell Technologies, Yokohama, Japan) was used to fabricate flexible PSCs. All fabrication processes were identical to that for BLP-PSC on FTO substrate, only the SnO₂ layer was annealed at 150 °C for 5 h.

2.5. Measurements

The surface morphologies were characterized using a field-emission scanning electron microscope (SEM) (S-4800, HITACHI, Tokyo, Japan). The photovoltaic characteristics were measured under 100 mW/cm² illumination using a solar simulator (Sol2A, Oriel, Irvine, CA, USA) with scan rate of 0.02 V at 25 °C. The internal electrochemical behavior was characterized using electrochemical impedance spectroscopy (EIS) (Compactstat.h, Ivium Technologies, Eindhoven, Netherlands) at a frequency range of 1 Hz to 1 MHz. The bending test was performed at a rate of 1 cycle per 0.5 s and a bending radius of 13 mm using a radius bending tester (JIRBT-620, JUNIL TECH, Daegu, Korea).

3. Results and Discussion

To investigate the possibility of using the conventional high-temperature-processed m-TiO₂ layer as the layer combined with the c-SnO₂ layer, we performed a comparative study of PSCs using both planar- and mesoporous-type PSCs. SL-PSCs in the configuration of FTO/c-SnO₂/perovskite/spiro-OMeTAD/Ag and BLH-PSCs in the configuration of FTO/c-SnO₂/m-TiO₂/spiro-OMeTAD/Ag were fabricated. In the case of the BLH-PSCs, the m-TiO₂ layer was sintered at 450 °C after spin-coating on the SnO₂ layer. The current density–voltage (J–V) curves under an irradiation of 100 mW cm^{−2} (AM 1.5) are shown in Figure 1. The SL-PSCs achieved a PCE of 13.68% with an open-circuit voltage (V_{OC}) of 0.99 V, a short-circuit current density (J_{SC}) of 19.77 mA/cm², and a fill factor (FF) of 70.22%. In contrast, the BLH-PSCs only achieved a PCE of 11.93% with a V_{OC} of 0.99 V, a J_{SC} of 19.52 mA/cm², and an FF of 61.86%. It is thus clear that SL-PSCs perform much better than BLH-PSCs. The large difference in FF could be primary attributed to the degradation of the SnO₂ layer by the high-temperature process [22].

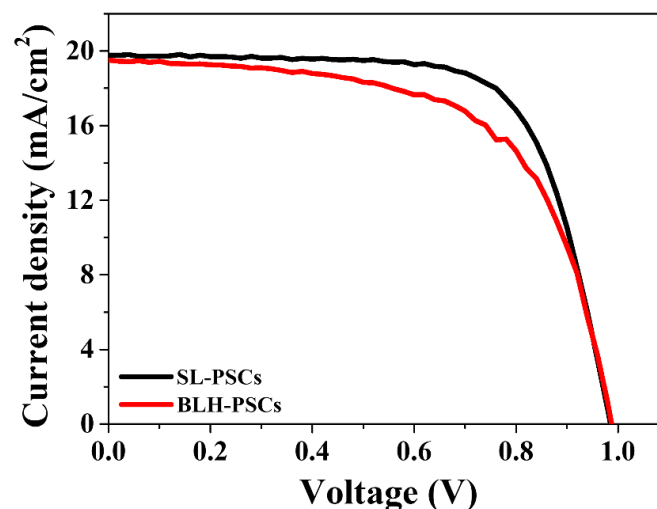


Figure 1. Reverse scan current density–voltage (J–V) curves of SL-PSC and BLH-PSC.

To uncover the underlying reasons for the decreased photovoltaic performance of BLH-PSCs, we fabricated SL-PSCs using a SnO₂ ETL annealed at temperatures from 200–500 °C. Figure 2 shows the dependence of PCE on the annealing temperature of the SnO₂ layer. As the annealing temperature increased, the photovoltaic performance of SL-PSCs decreased. The PCEs of SL-PSCs annealed at 200, 300, and 400 °C were 13.68%, 12.19%, and 8.90%, respectively. The SL-PSCs annealed at 400 °C performed poorly, with very low FF and J_{SC}. Moreover, the SL-PSCs annealed at 500 °C did not show any photovoltaic characteristics. The detailed photovoltaic parameters obtained from the J-V curves are summarized in Table S1.

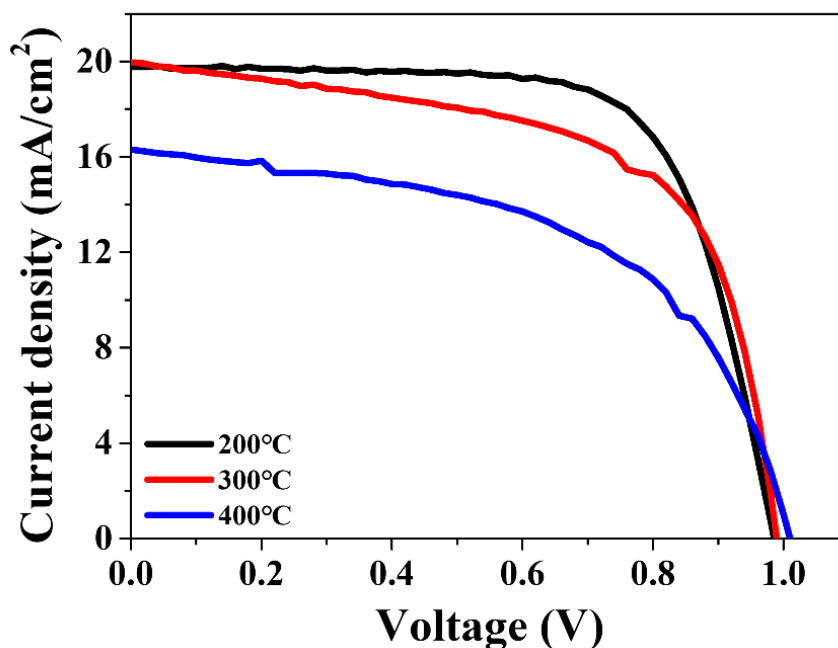


Figure 2. Reverse scan current density–voltage (J-V) curves of perovskite solar cells based on SnO₂ ETL annealed at 200, 300, and 400 °C.

To verify the decreasing trend in PCE, we investigated the morphology change of the SnO₂ layer according to the annealing temperature. Figure 3 shows the top-view SEM images of SnO₂ layers deposited on FTO substrates and annealed at different temperatures. As shown in Figure 3a, the FTO substrate was uniformly covered with the SnO₂ layer, and no pinholes were observed when the SnO₂ layer was annealed at the relatively low temperature of 200 °C. However, as the annealing temperature increased above 300 °C, the SnO₂ nanoparticles agglomerated more and the FTO areas uncovered by SnO₂ increased. The high-temperature-annealed SnO₂ layer could not completely cover the FTO substrate, and thus these pinholes resulted in leakage in the current pathway. Moreover, poor interface contact with the FTO substrate increased the series resistance. Therefore, as shown in Figure 1, BLH-PSCs exhibited lower FF and PCE than SL-PSCs because the high-temperature process of the m-TiO₂ layer caused the degradation of the underlying SnO₂ layer [24]. The above results confirm that the low-temperature processing of the m-TiO₂ layer without causing damage to the SnO₂ layer is important for producing high-performance PSCs using a c-SnO₂/m-TiO₂ bilayer ETL.

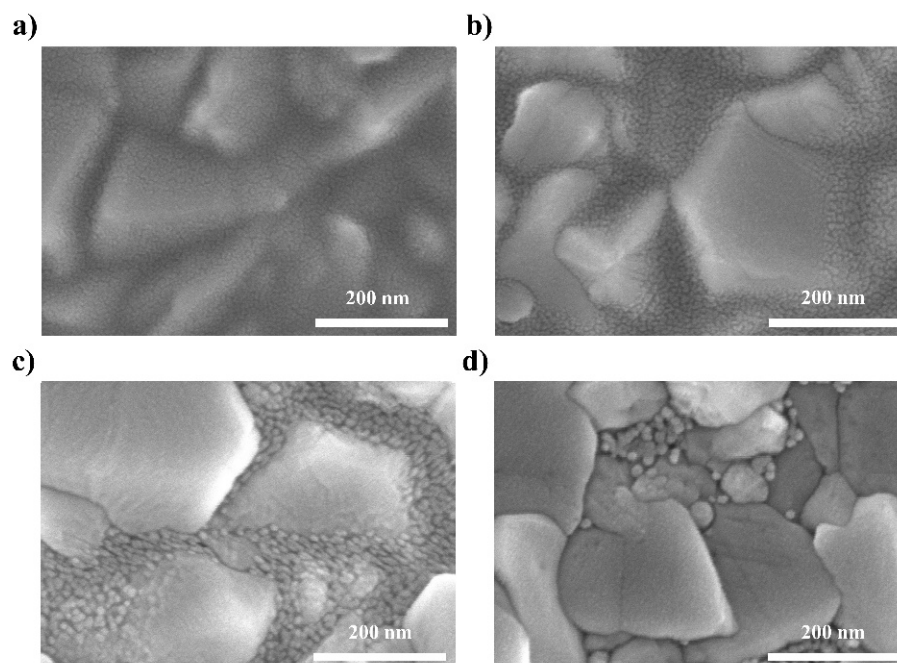


Figure 3. Top SEM image of SnO₂ layer after annealing at (a) 200, (b) 300, (c) 400, and (d) 500 °C on FTO glass.

To investigate the effectiveness of our strategy for LT m-TiO₂, we compared the PCEs of three different PSCs (Figure S2). First, we fabricated SL-PSC without the m-TiO₂ layer to determine the role of the m-TiO₂ layer. Then, we fabricated two different PSCs based on the FTO/c-SnO₂/m-TiO₂/spiro-OMeTAD/Ag architecture. The main difference between these two PSCs with a bilayer ETL was the post-treatment of the m-TiO₂ layer. In the case of LT m-TiO₂-based PSCs (BLL-PSCs), the m-TiO₂ layers were only annealed at 150 °C, whereas oxygen plasma treatment was directly performed on the LT TiO₂ layers for BLP-PSCs. As shown in Figure 4, the BLL-PSCs exhibited a PCE of only 5.43% with a V_{OC} of 0.85 V, a J_{SC} of 14.81 mA/cm², and an FF of 43.01% (Table 1). Although the low-temperature processing of m-TiO₂ might not have caused the aggregation of the underlying c-SnO₂ layer, the remaining organic additives in m-TiO₂ inhibited the full coverage of the perovskite layer on TiO₂, which hindered electron transport at the interface between the perovskite and TiO₂. According to our previous work, oxygen plasma treatment can successfully remove organic additives from and improve the wettability of the LT TiO₂ layer [13]. With oxygen plasma treatment, the performance of BLL-PSCs considerably improved, and the PCE, V_{OC}, J_{SC}, and FF were 15.35%, 1.03 V, 20.65 mA/cm², and 72.30%, respectively. Moreover, the PCE of BLP-PSC (15.53%) is higher than that of SL-PSC (13.68%). These results demonstrate that oxygen plasma treatment enables the fabrication of c-SnO₂/m-TiO₂ bilayer ETL-based PSCs with excellent photovoltaic performance using an LT m-TiO₂ layer.

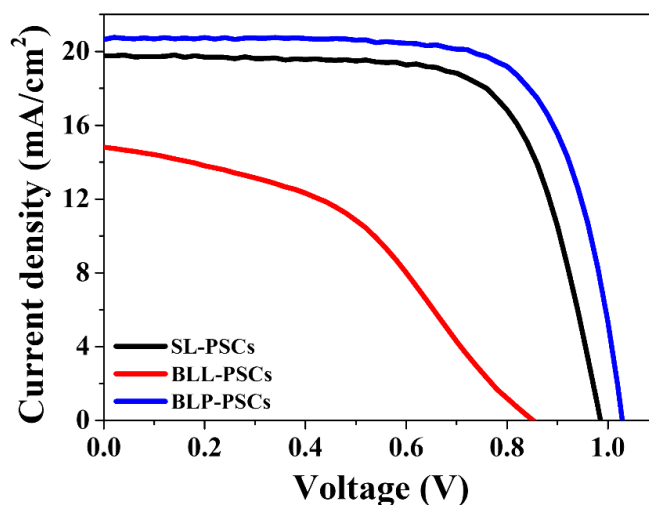


Figure 4. Reverse scan current density–voltage (J-V) curves of SL-PSCs, BLL-PSCs, and BLP-PSCs.

Table 1. Summary of the photovoltaic parameters of PSCs based on different ETLs.

Sample	J_{SC} (mA/cm ²)	V_{OC} (V)	FF (%)	PCE (%)
SL-PSCs	19.77	0.99	70.22	13.68
BLL-PSCs	14.81	0.85	43.01	5.43
BLP-PSCs	20.65	1.03	72.30	15.36

To gain further insight into the effects of the m-TiO₂ layer on charge transfer properties at the ETL/perovskite interface, EIS was conducted. The Nyquist plots of different ETLs were obtained in the dark with an applied bias voltage of 0.9 V and are shown in Figure 5. The series resistance (R_s) and charge transport resistance (R_{ct}) were obtained by fitting EIS data according to the relevant equivalent circuit, as shown in the inset of Figure 5. The EIS parameters from the semicircle Nyquist plot are summarized in Table S2. In general, R_s is related to the sheet resistance of electrodes [25], including the contributions from FTO and metal electrodes. In contrast, R_{ct} generally refers to the charge transfer resistance at all the interfaces [26], such as between the carrier selective layer and the perovskite layer, and between the electrode and the carrier selective layer. The R_{ct} value of SL-PSC was 350 Ω , which is slightly higher than that of BLP-PSC (230 Ω). The small R_{ct} value of BLP-PSC further supports that the combination of the m-TiO₂ layer with c-SnO₂ promotes good interface contact between the ETL and perovskite, leading to an enhanced charge transfer process and the highest PCE [27]. While BLL-PSC exhibited the highest value of R_{ct} , it had the lowest PCE. Figure S1 shows the top-view SEM images of the perovskite layer on c-SnO₂ and c-SnO₂/m-TiO₂ and their corresponding grain size distribution histograms. Both perovskite layers exhibited a similar average grain size with uniform morphology consisting of densely packed grains. Because the grain size, which refers to the density of grain boundaries, is related to the transport of photogenerated carriers and the extension of the charge carrier diffusion length [28], a comparable grain size might not result in different photovoltaic characteristics. However, the mesoporous structure of TiO₂ allows the perovskite to infiltrate into TiO₂, which increases the interfacial contact between perovskite and TiO₂ (Figure S3). Therefore, the improved interfacial contact due to the direct transfer pathway substantially contributed to the increase in PCE of SL-PSCs by combination with the m-TiO₂ ETL (Figure S4 and Table S3).

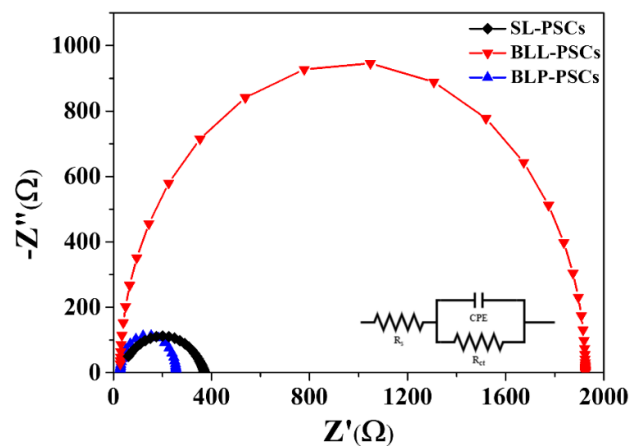


Figure 5. Nyquist plots of SL-PSCs, BLL-PSCs, and BLP-PSCs.

The high performance of c-SnO₂/m-TiO₂ bilayer ETL-based PSCs was achieved with oxygen plasma treatment at a low temperature, which suggests that high-performance and flexible BLP-PSCs can also be attained via the same procedures. Figure 6 shows the J-V curves of the flexible BLP-PSCs constructed on the PEN/ITO substrate as a function of bending cycles. The flexible cells exhibited a promising PCE of 9.56%, with a V_{OC} of 1.81 V, a J_{SC} of 17.20 mA/cm², and an FF of 55.22%. The inferior performance of the flexible cell compared to the rigid cell on the FTO/glass substrate arose from its inferior surface morphology and low transmittance compared to the rigid surface [29]. To investigate the mechanical stability of the flexible cell, a bending durability test was performed. All the photovoltaic characteristics were maintained during 500 cycles of bending without much deterioration.

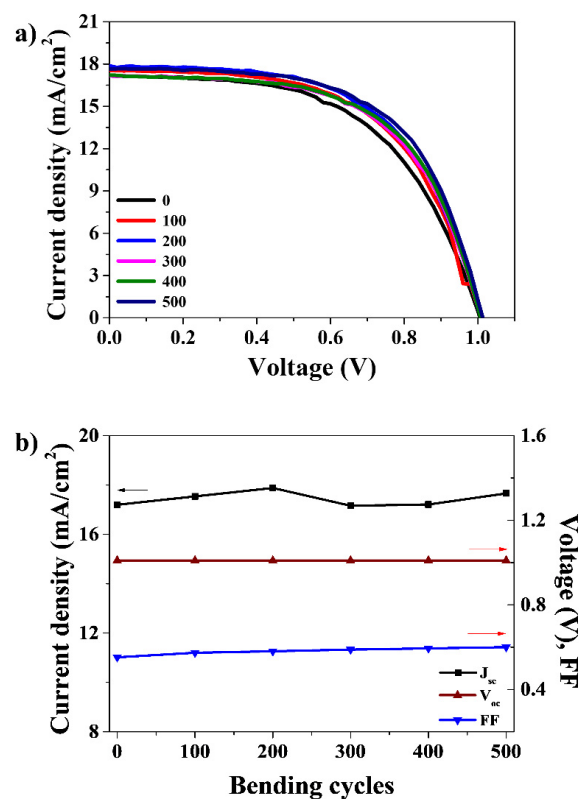


Figure 6. (a) Reverse scan current density–voltage (J-V) curves of flexible cells and (b) variation in the J_{SC}, V_{OC}, and FF values of flexible cells as a function of bending cycles.

4. Conclusions

In summary, an LT m-TiO₂ layer using oxygen plasma treatment was combined with c-SnO₂. The BLP-PSC had a PCE of 15.36%, which is much higher than that of the PSC with a single c-SnO₂ ETL. This high efficiency was obtained because the oxygen plasma treatment facilitated the removal of organic additives from LT m-TiO₂ and the infiltration of perovskite into m-TiO₂, thus enhancing charge transport and extraction. This proves that our strategy to construct a bilayer ETL using LT m-TiO₂ is beneficial because it allows the superior characteristics of the underlying c-SnO₂ to be maintained, and the mesoporous structure provided increased interfacial contact between the perovskite and the ETL. Consequently, the c-SnO₂/m-TiO₂ bilayer ETL is thought to be one of the most promising ETL layers for high-efficiency PSCs.

Supplementary Materials: The following supporting information can be downloaded at: <https://www.mdpi.com/article/10.3390/nano12040718/s1>, Figure S1: Top-view SEM images and grain size distribution histograms (inset) of MAPbI₃ layer on SL, BLL, BLP, and BLH; Figure S2: XRD pattern of MAPbI₃ layer on BLP, BLL, and SL; Figure S3: Cross-section SEM image of glass/FTO/ETL/MAPbI₃; Figure S4: Reverse scan (solid line) and forward scan (dotted line) current density–voltage curves of PSCs based on SL-PSC, BLL-PSC, and BLP-PSC; Table S1: Summary of the photovoltaic parameters of PSCs based on SnO₂ ETL annealed at 200, 300, and 400 °C; Table S2: Summary of EIS data; Table S3: Summary of the photovoltaic parameters of SL-PSC, BLL-PSC, and BLP-PSC according to scan direction.

Author Contributions: S.J. conceived and designed the research; J.L. participated in material preparation and device fabrication; J.K. and C.-S.K. participated in data interpretation; S.J. and J.L. wrote the paper; S.J. supervised the project. All authors have read and agreed to the published version of the manuscript.

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Data Availability Statement: The data presented in this article is available on request from the corresponding author.

Conflicts of Interest: The authors declare no conflict of interest.

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